

Colours from Earth

an inward perspective

Bob Fosbury (The^Bob)
@
Villa Aureli, May 2011

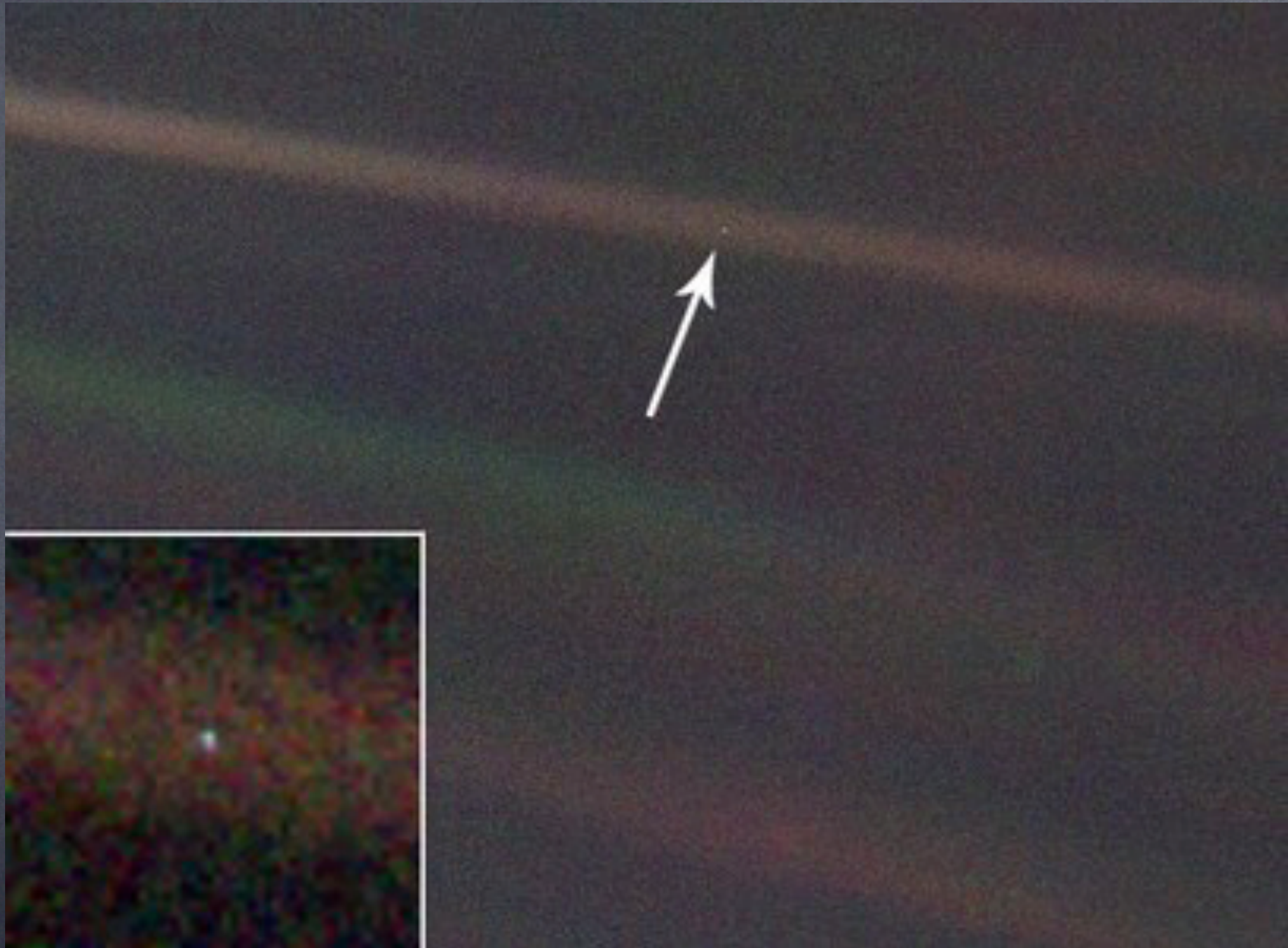


Colours from Earth

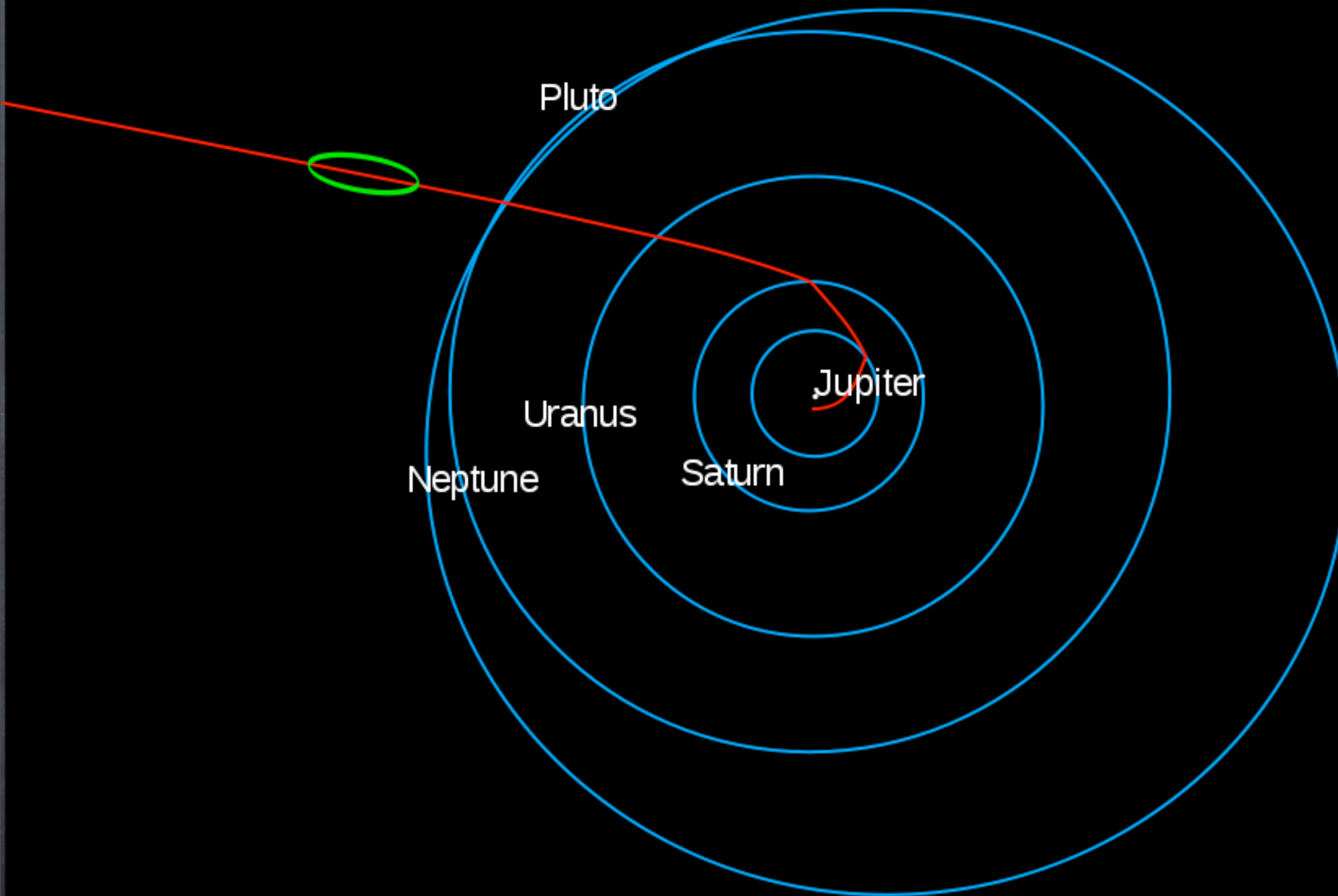
an inward perspective

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The "Pale Blue Dot"

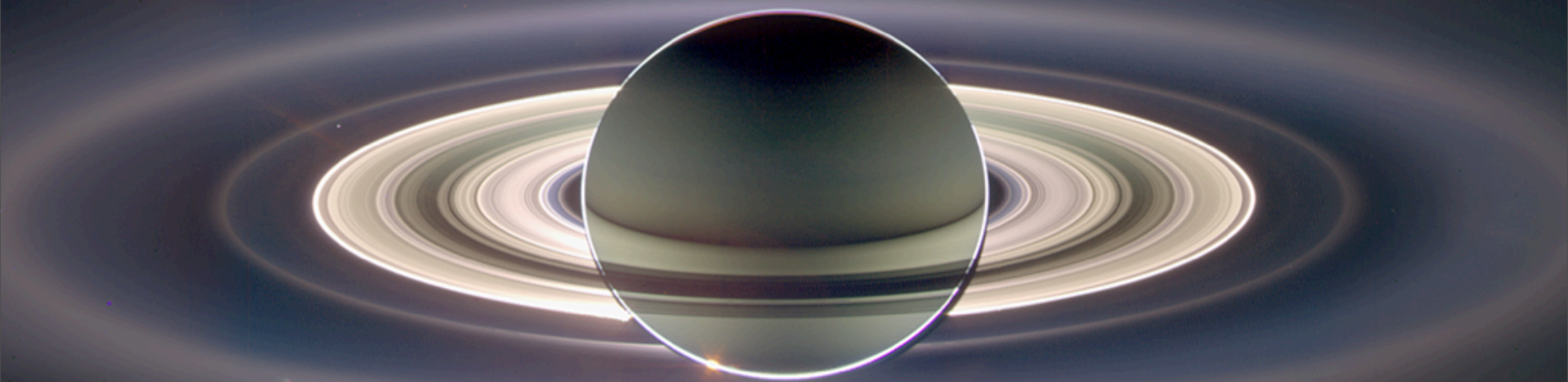


...taken in 1990 by Voyager 1 from 6.1 billion km



...taken in 1990 by Voyager 1 from 6.1 billion km

From behind Saturn



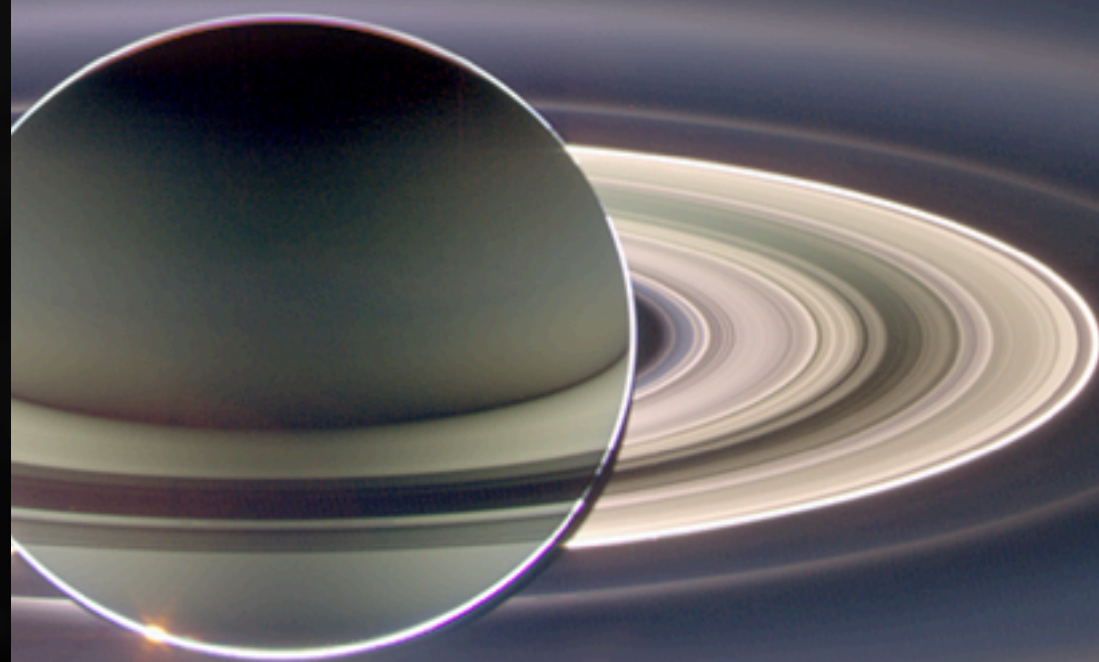
Cassini Wide Angle Camera, 15 Sep. 2006

From behind Saturn



Cassini Wide Angle Camera, 15 Sep. 2006

From behind Saturn



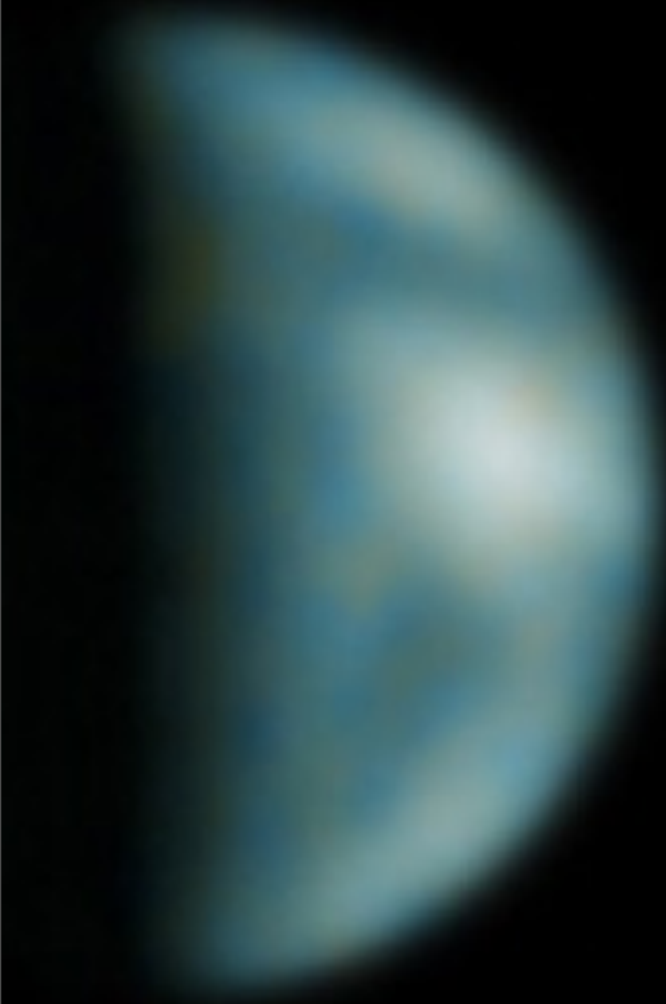
Cassini Wide Angle Camera, 15 Sep. 2006

From a neighbour

Mars Global Surveyor
Mars Orbiter Camera
on
8 May 2003
from 139 million km



From a neighbour



Earthrise



... from the Moon: Apollo 8, December 1968

From Earth orbit



Animation by Martin Kornmesser

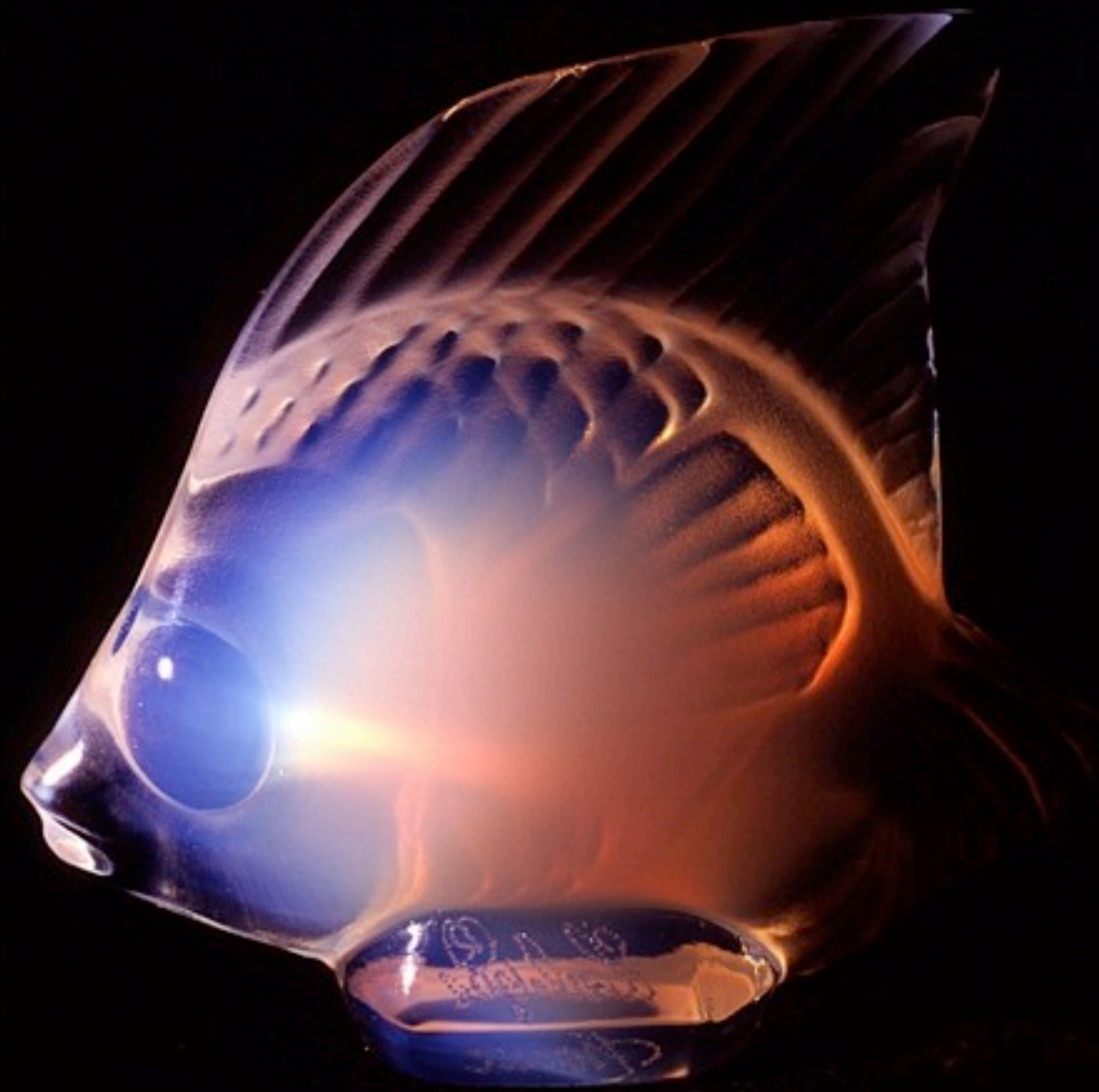
The "Blue Planet"
The "Pale Blue Dot"

Why blue?

The simple answer: because of air and water
But we can look a bit deeper...

The "Blue Planet"
The "Pale Blue Dot"

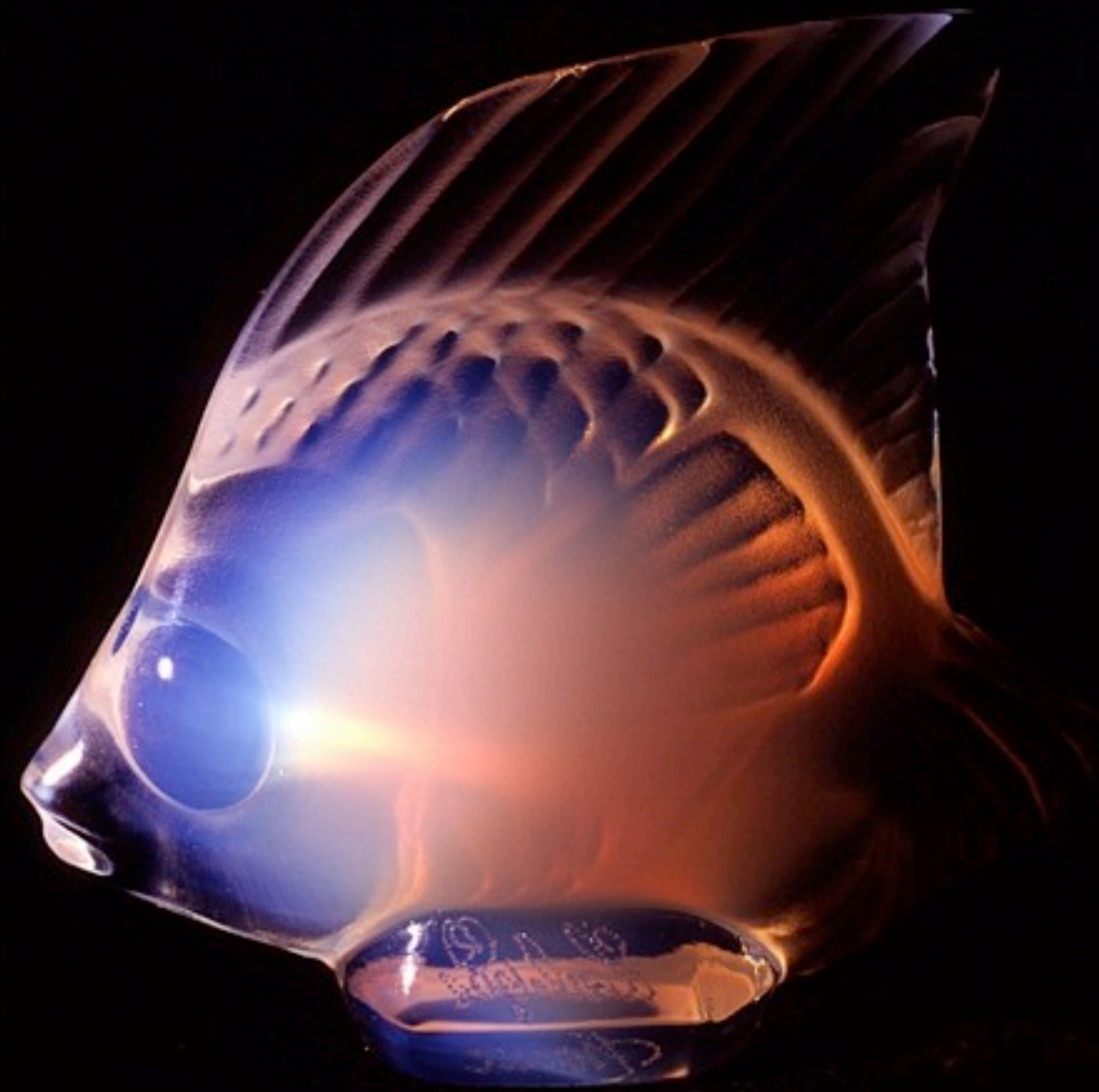
Why blue?



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The "Blue Planet"
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Why blue?



The simple answer: because of air and water
But we can look a bit deeper...

Water is blue

Unlike many blues in nature, water is actually blue

Water molecules absorb red light; in fact water absorbs pretty much all electromagnetic radiation except blue light and metre-length radio waves

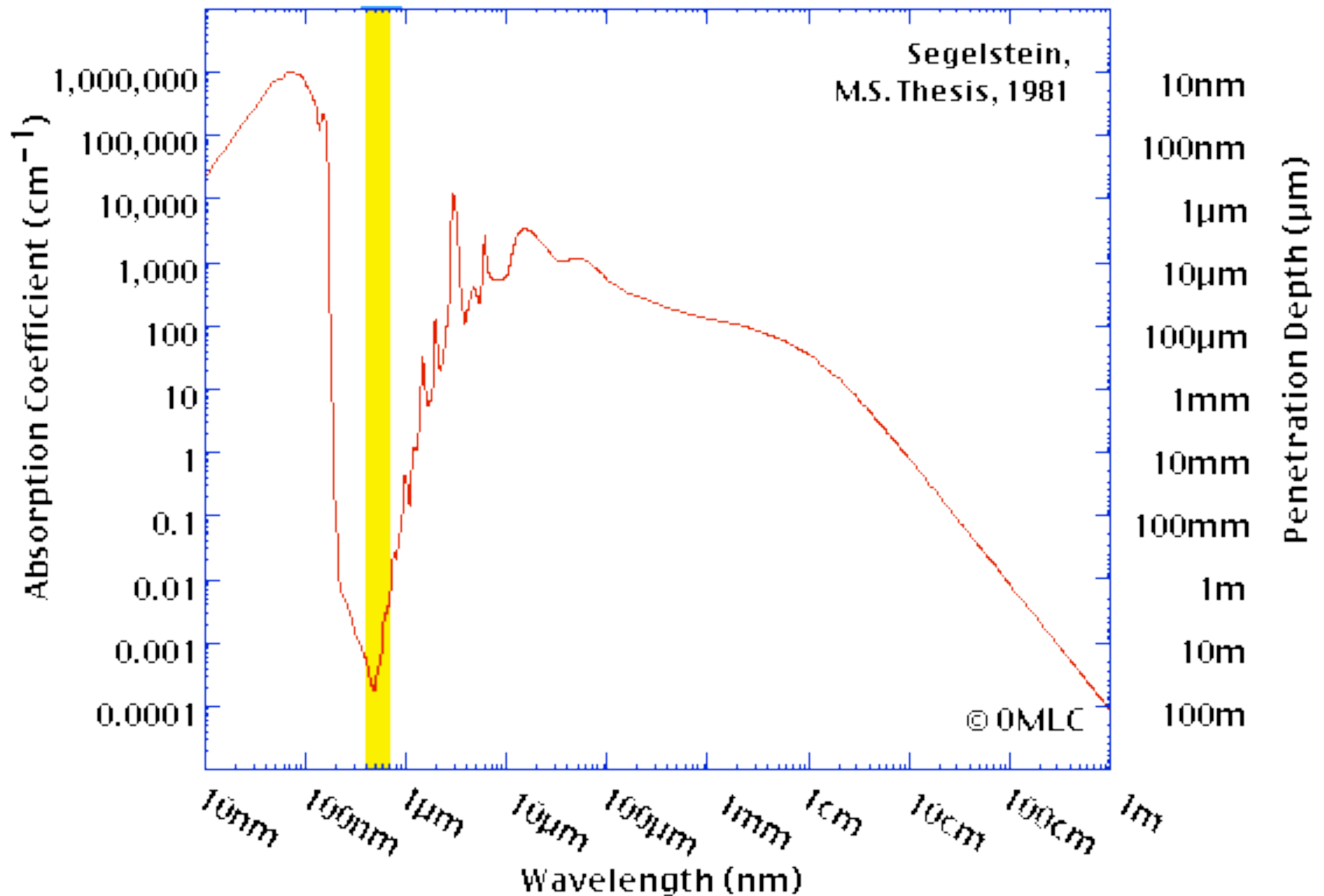
Very clear seawater on the Cornish coast





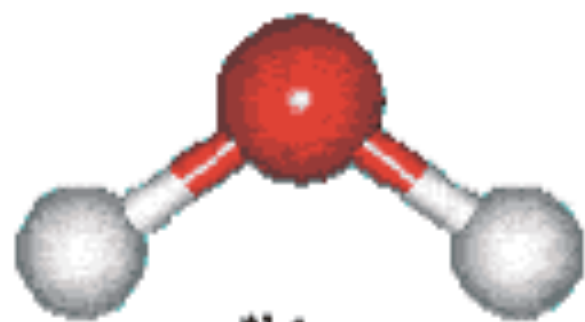
Iceberg, Glacier Bay, Alaska

The “Water hole”



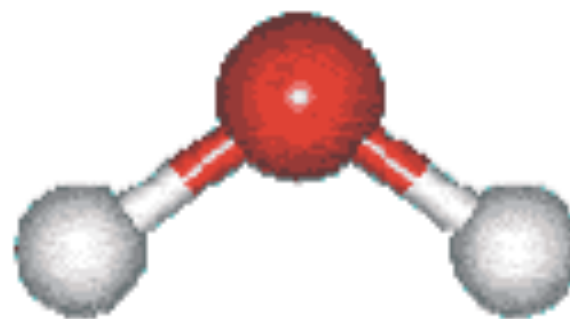
Assignment of the IR vibrational absorption spectrum of liquid water

Wavelength	cm ⁻¹	Assignment	Wavelength	cm ⁻¹	Assignment
0.2 mm	50	intermolecular bend	1470 nm	6800	$av_1 + bv_3; a+b=2$
55 μm	183.4	intermolecular stretch	1200 nm	8330	$av_1 + v_2 + bv_3; a+b=2$
25 μm	395.5	L ₁ , librations	970 nm	10310	$av_1 + bv_3; a+b=3$
15 μm	686.3	L ₂ , librations	836 nm	11960	$av_1 + v_2 + bv_3; a+b=3$
6.08 μm	1645	v ₂ , bend	739 nm	13530	$av_1 + bv_3; a+b=4$
4.65 μm	2150	v ₂ + L ₂ ^b	660 nm	15150	$av_1 + v_2 + bv_3; a+b=4$
3.05 μm	3277	v ₁ , symmetric stretch	606 nm	16500	$av_1 + bv_3; a+b=5$
2.87 μm	3490	v ₃ , asymmetric stretch	514 nm	19460	$av_1 + bv_3; a+b=6$
1900 nm	5260	$av_1 + v_2 + bv_3; a+b=1$	Note that a and b are integers, ≥ 0 ms.		



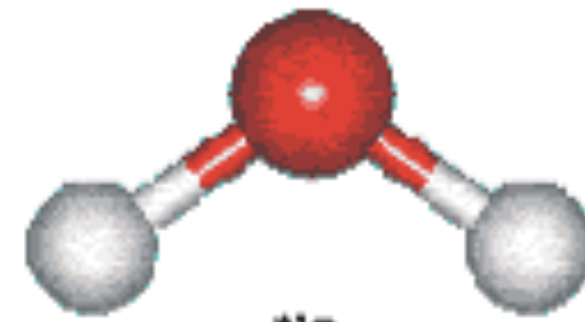
ν_1

symmetric stretch



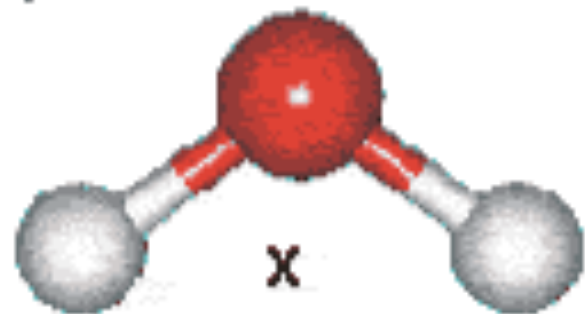
ν_3

asymmetric stretch



ν_2

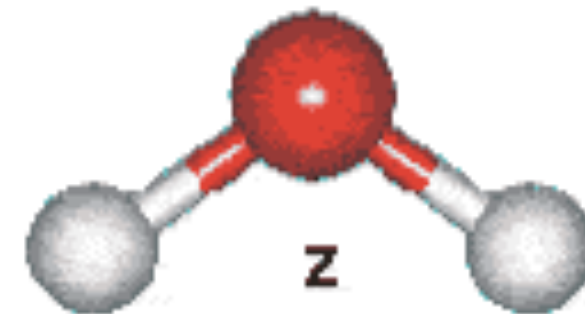
bend



x



y



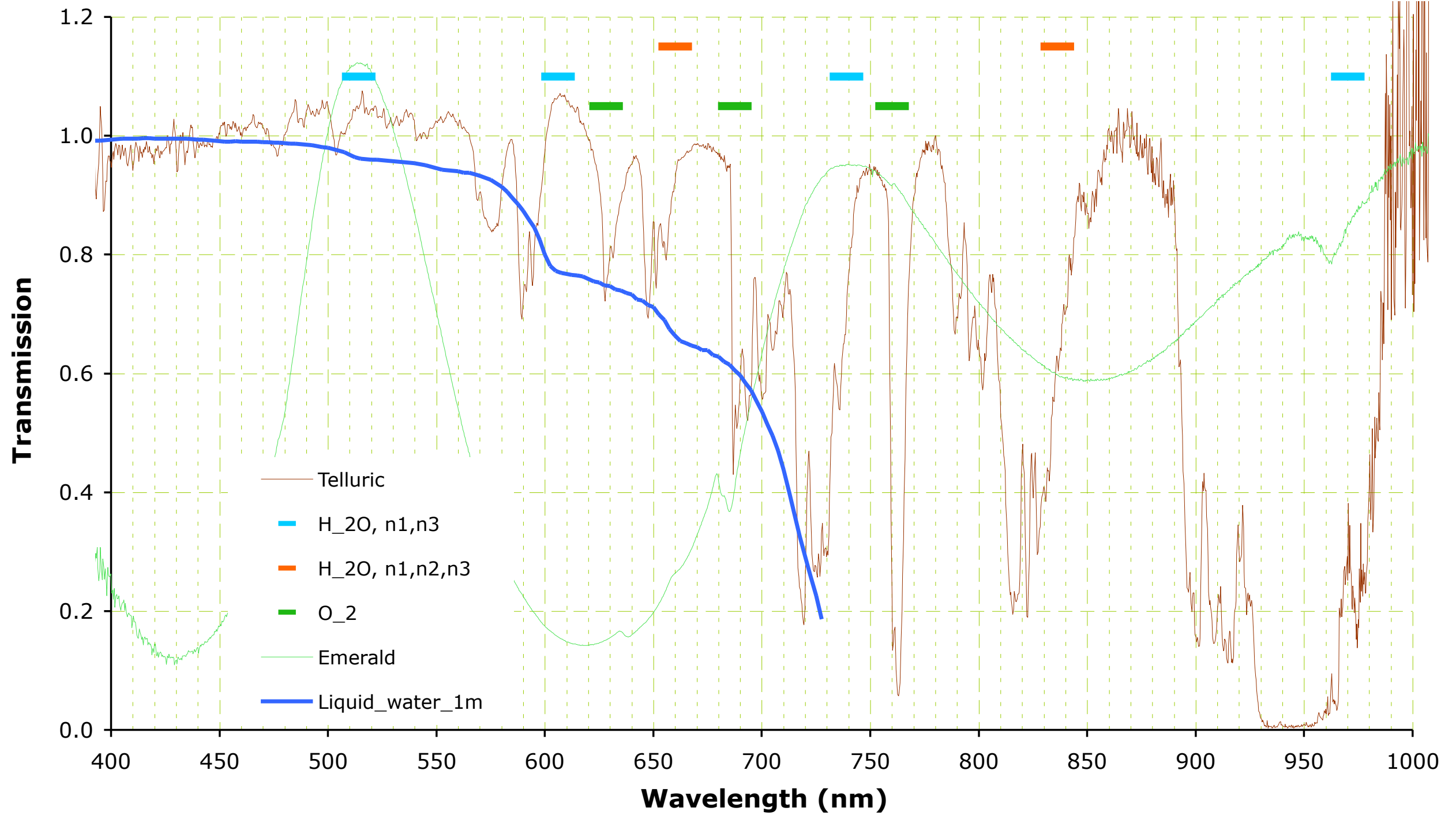
z

librations

Assignment of the IR vibrational absorption spectrum of liquid water

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Water absorption in the atmosphere at sunset, in 1m of liquid and in an emerald crystal







The blue colour of heavily overcast skies

Lee et al. 2005

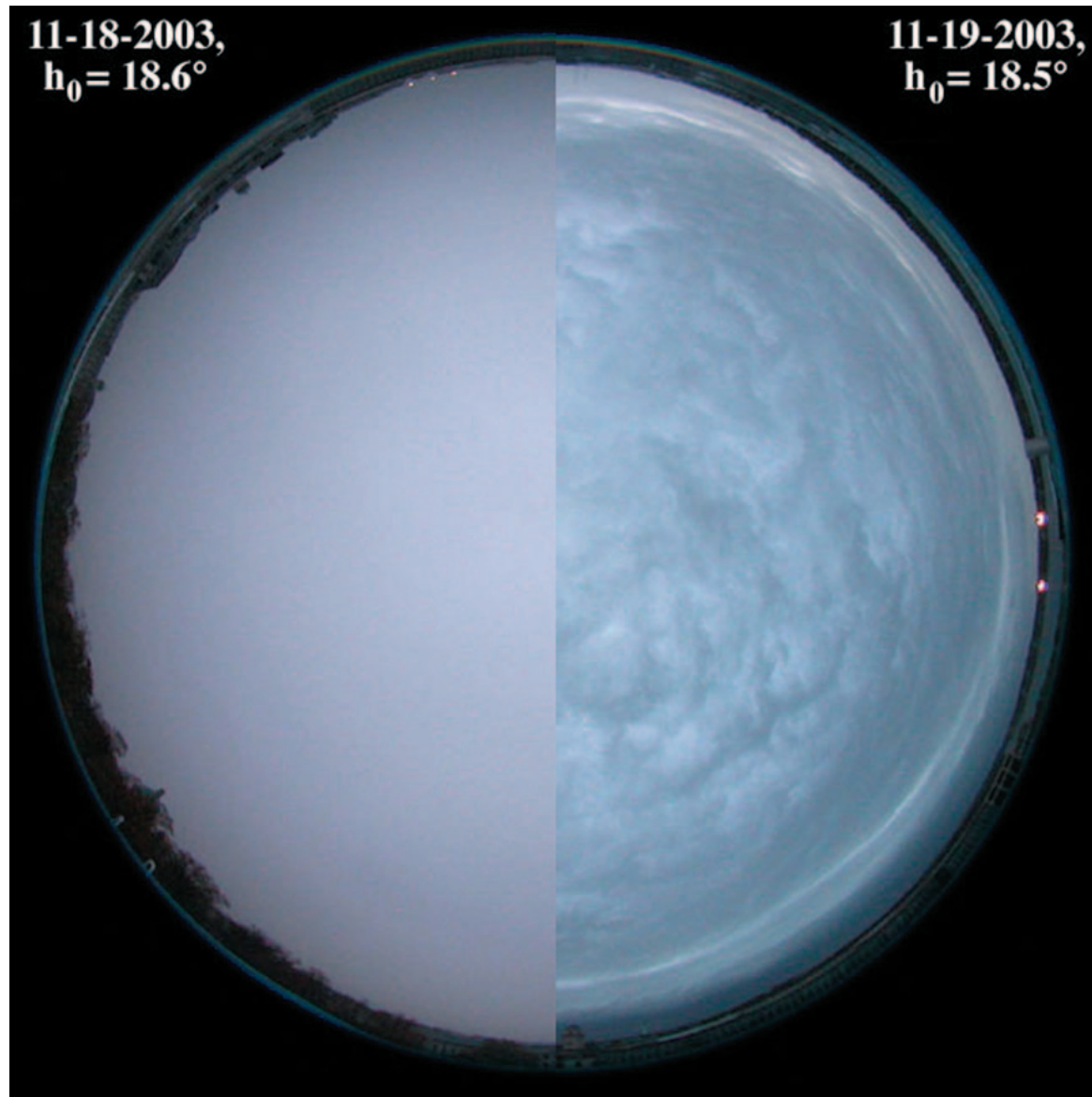


Fig. 5. Composite fisheye image of a bright stratus overcast (left half, $E = 24.3 \text{ W m}^{-2}$) and a much darker stratocumulus overcast (right half, $E \sim 3.6 \text{ W m}^{-2}$) at USNA on 11-18-03 and 11-19-03, respectively. Note that the darker overcast is distinctly bluer. Although exposures differ, in both original photographs the digital camera's white-balance setting was the same and $h_0 \sim 18.5^\circ$.

Mostriamo in figura 4.1 il grafico della distribuzione del numero di interazioni

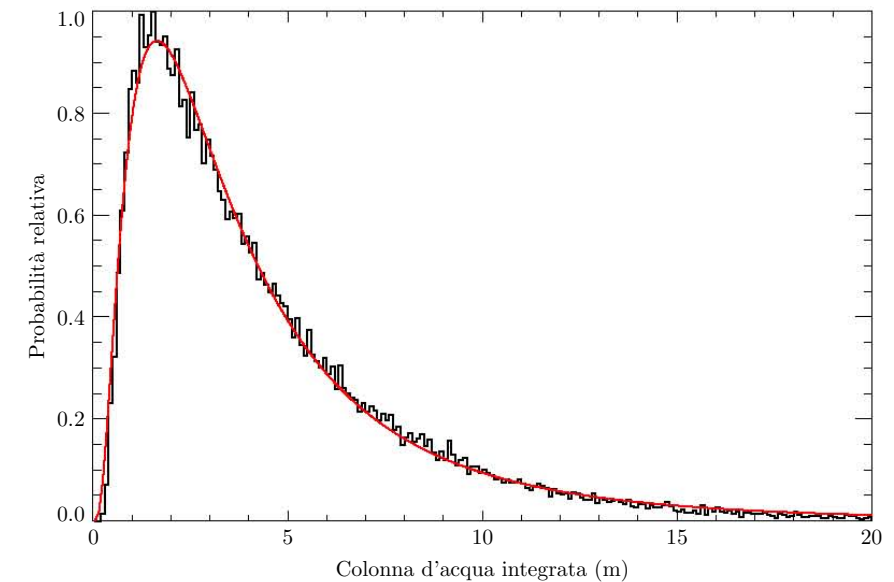


Figura 4.1: Istogramma dei valori della colonna d'acqua. In rosso è rappresentata la curva di fit con andamento log-normale.

Il numero di conteggi inizialmente presente sull'ordinata è stato sostituito con una distribuzione di probabilità normalizzata sul valore di conteggi maggiore (posto quindi uguale a 1). Riportiamo i valori della colonna d'acqua ottenuti dalla simulazione:

$$W \text{ (valor medio)} = 4.38 \text{ m} ,$$

$$W \text{ (mediana)} = 3.32 \text{ m} ,$$

$$\sigma = 2.77 \text{ m} . \quad (4.1)$$

il fit migliore dell'istogramma è dato dalla curva log-normale; la distribuzione log-normale con cui abbiamo fittato l'istogramma presenta una sigma molto alta, cosa che del resto si riscontra nella σ dei nostri dati (eq. 4.1).

Plot for 1 cm precipitable liquid water

Air is blue

Yes, we all know the clear sky is blue because of Rayleigh scattering from molecules and density fluctuations, but air is actually blue as well!

This is because of ozone, the unstable allotrope of oxygen, which exists mostly between 15 and 40 km altitude in the atmosphere

Rayleigh scattering blue

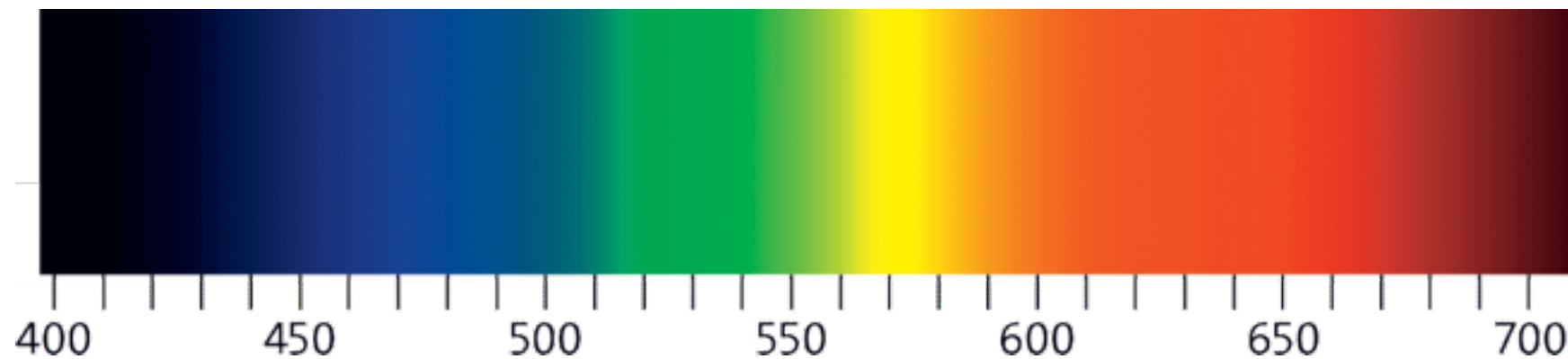
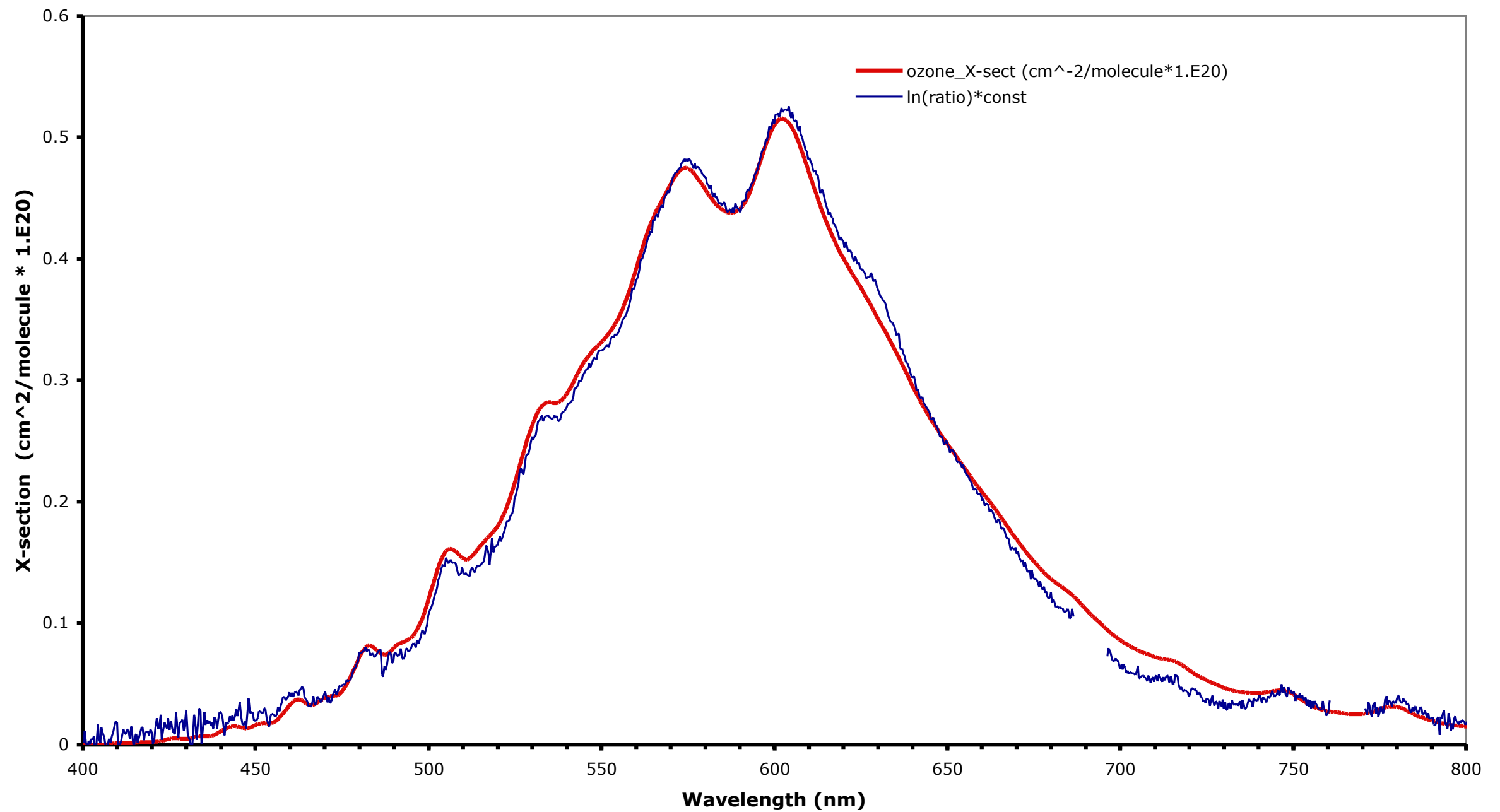


Rayleigh scattering blue

Mie scattering
white



The Chappuis band of ozone



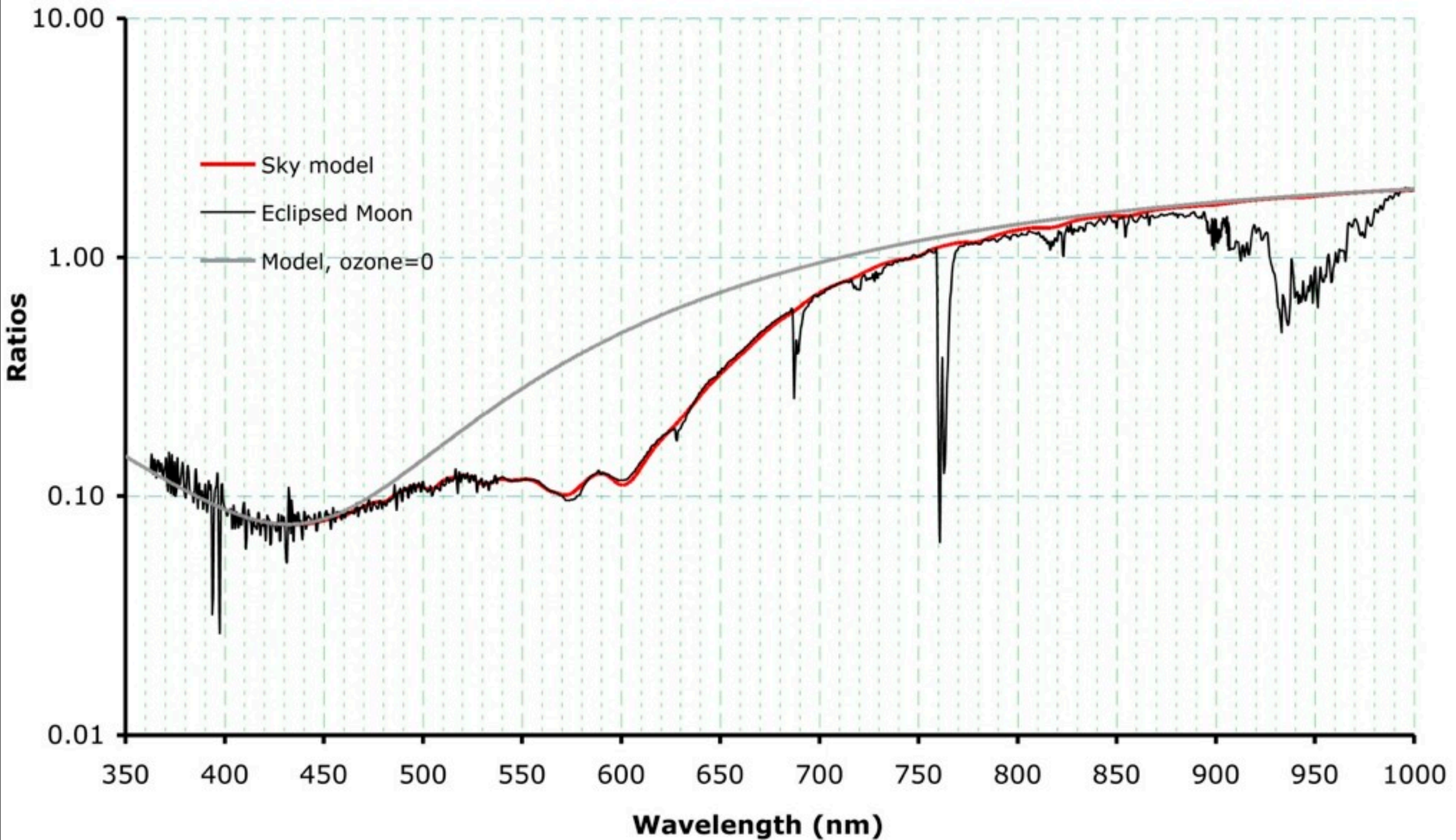


This is the sky seen by the eclipsed Moon
at the boundary of the umbra/penumbra

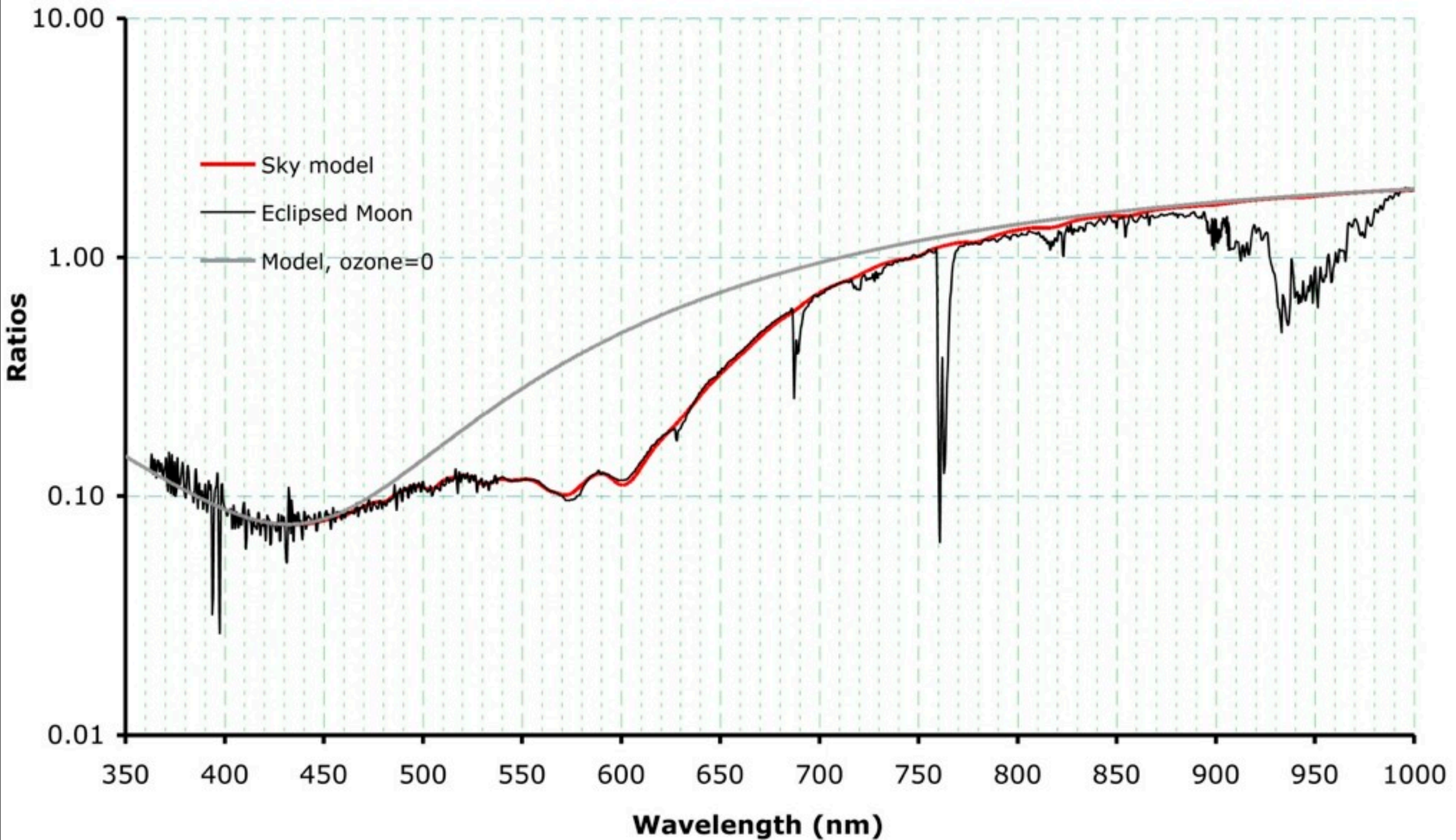


Thanks to Zolt Levay

Lunar eclipse, 16 August 2008



Lunar eclipse, 16 August 2008

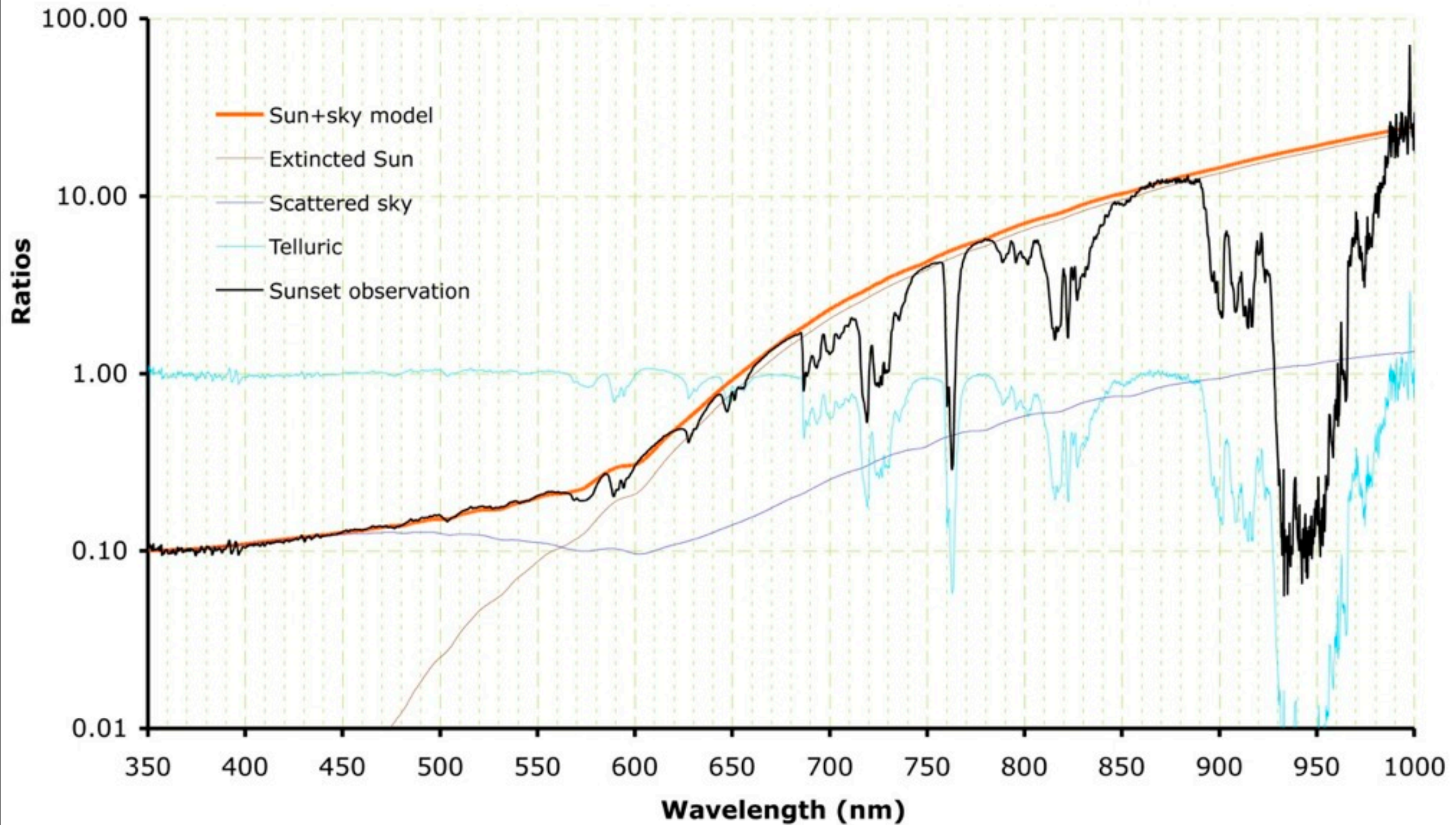


A landscape photograph showing a sunset over a green field. The sun is a small red circle on the horizon, partially obscured by a dark line of trees. The sky is filled with horizontal bands of orange, pink, and blue clouds. A large white circle is superimposed over the entire scene, centered on the sun. The text "Spectrometer input aperture" is written in white at the bottom of the circle.

Spectrometer input aperture

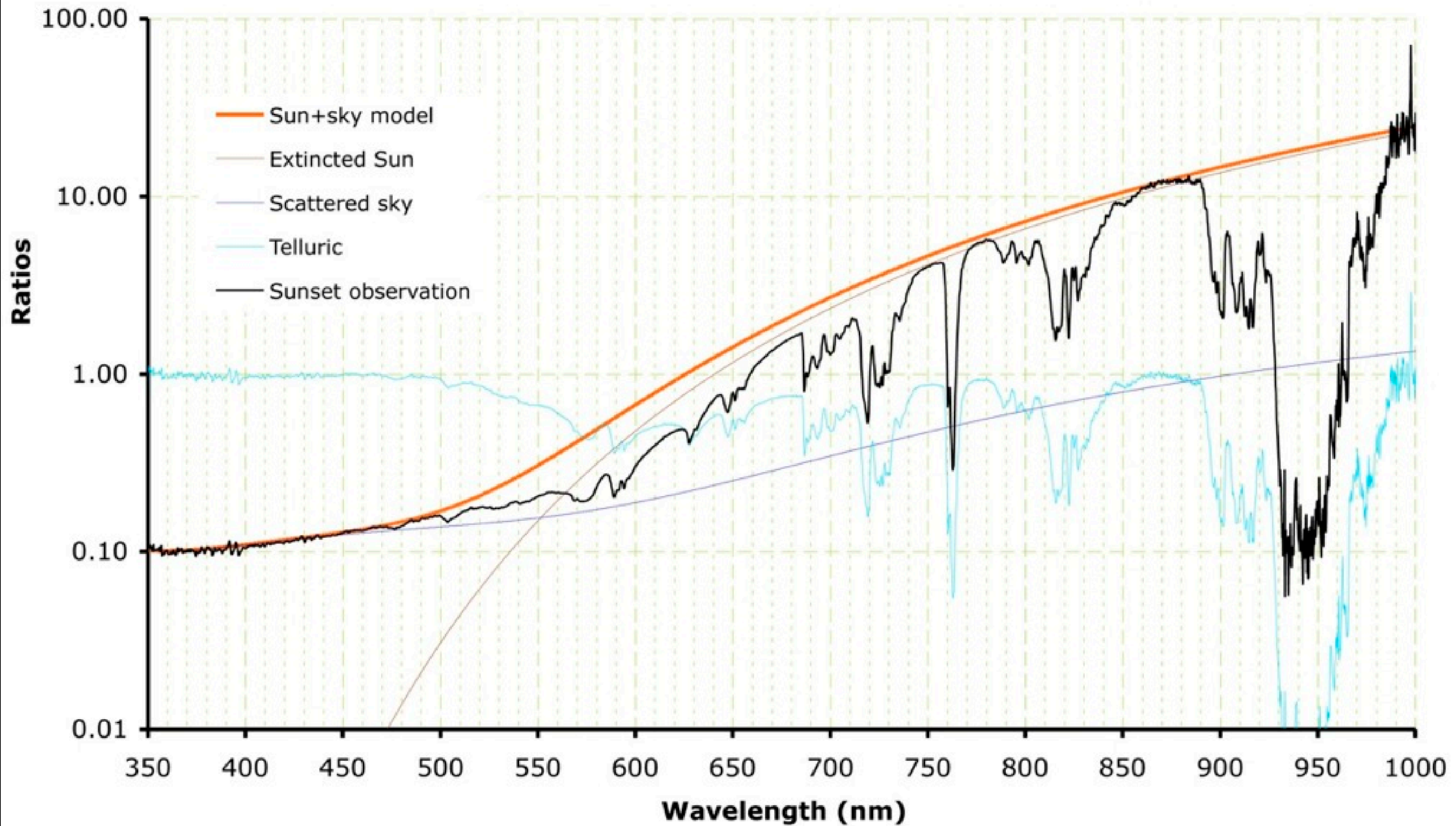
Sunset spectra

Sunset, 27 June 2010, Norfolk, England



Sunset spectra

Sunset, 27 June 2010, Norfolk, England; ozone=0



Sunset sky (model)

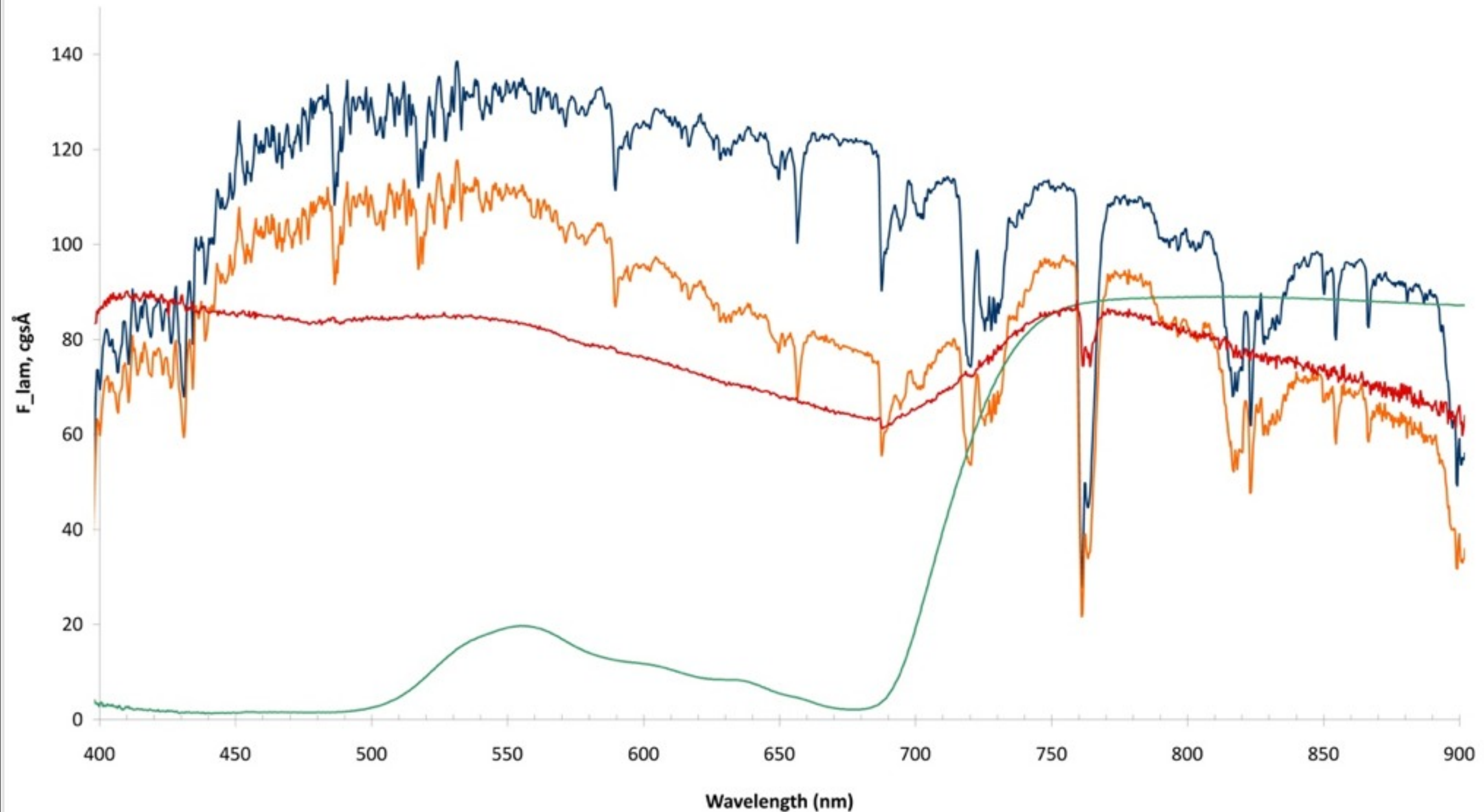
No ozone

The Green Planet

Chlorophyll in action

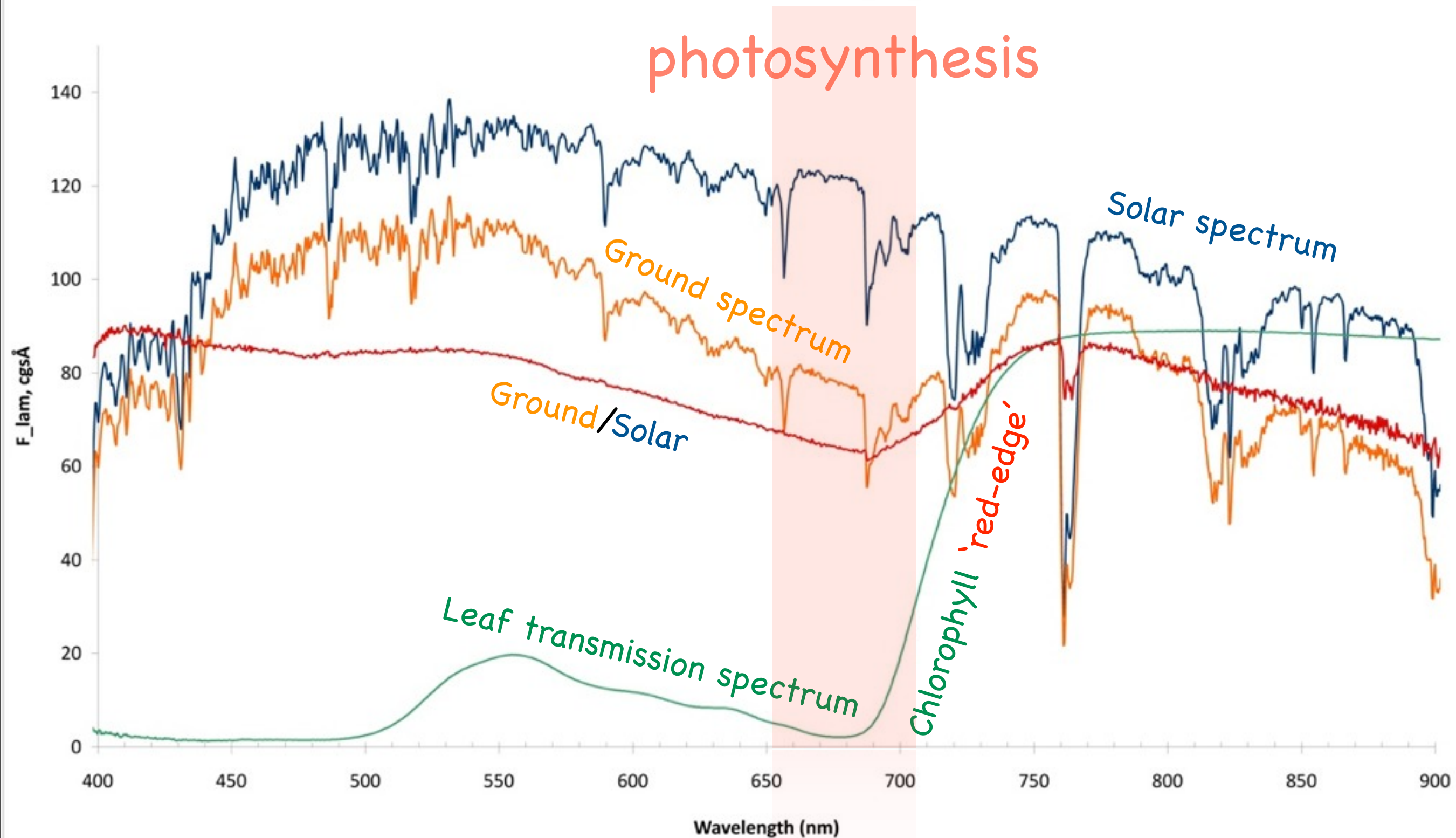


Daylight ground overcast, Sun $z_d=57^\circ$



Spectrum of the ground: mixed buildings, grass and trees
Seeing the **chlorophyll** 'red-edge'

Daylight ground overcast, Sun zd=57°



Spectrum of the ground: mixed buildings, grass and trees
Seeing the **chlorophyll 'red-edge'**



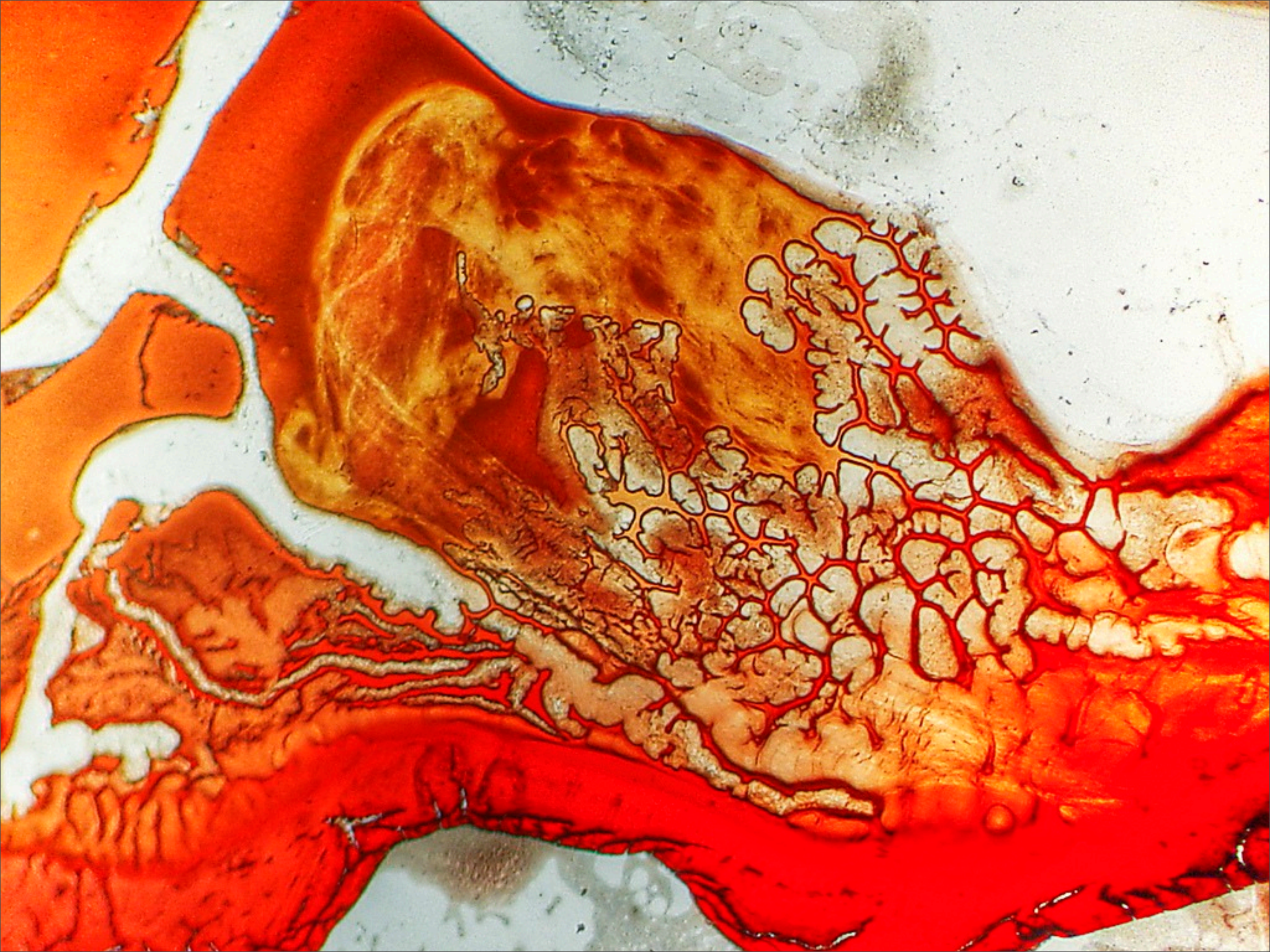
Spectrum of the ground: mixed buildings, grass and trees
Seeing the **chlorophyll** 'red-edge'

Natural colours on Earth

Chlorophyll, like haemoglobin, is a 'porphyrin' and there are other examples of this class that produce colours we see

The brown of a hen's eggshell is actually a white matrix speckled with purple protoporphyrin and, like chlorophyll, fluoresces red under UV light

Seaweeds contain auxiliary pigments that transfer energy from underwater (mostly blue!) photons to the photosynthetic mechanisms in chlorophyll



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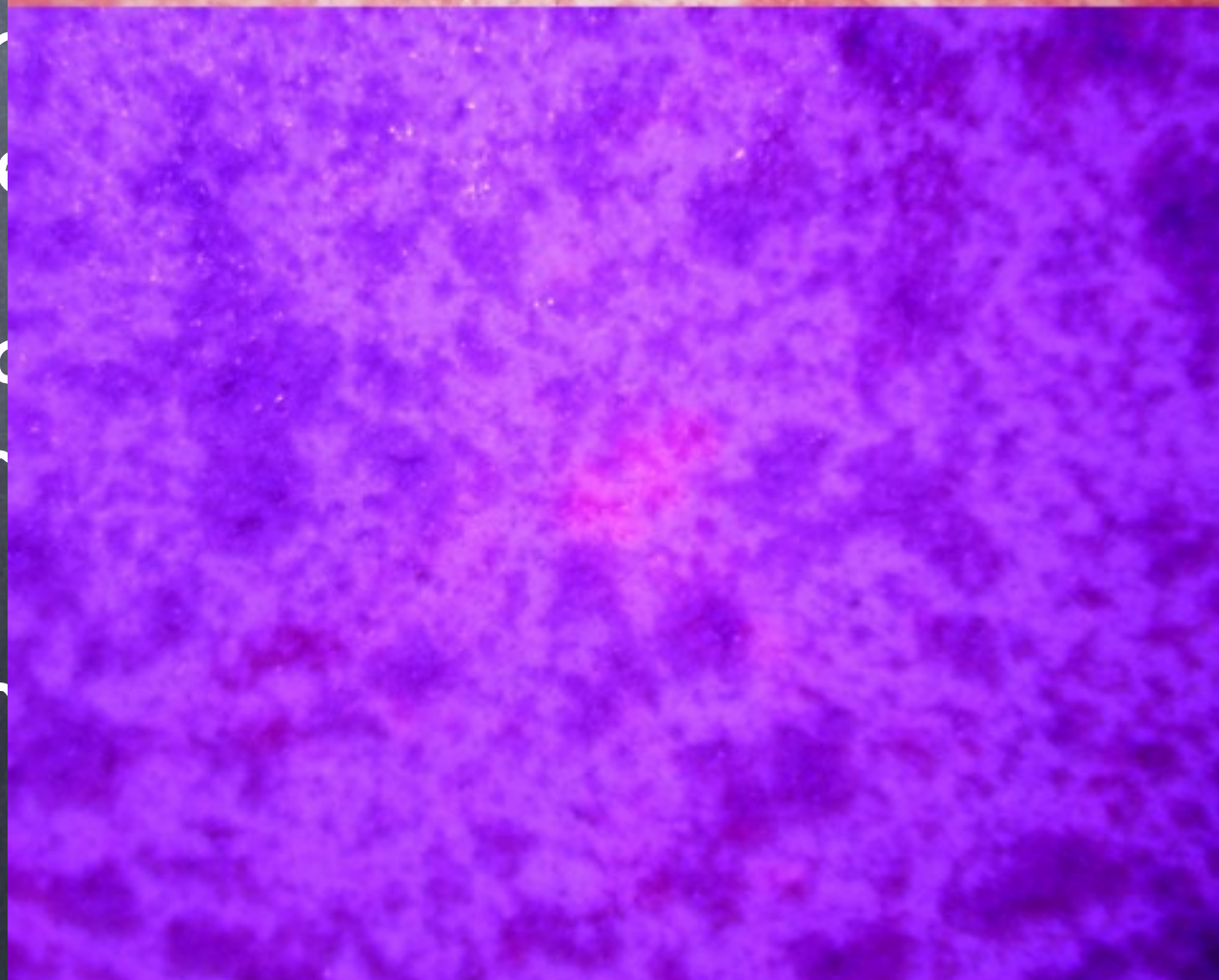
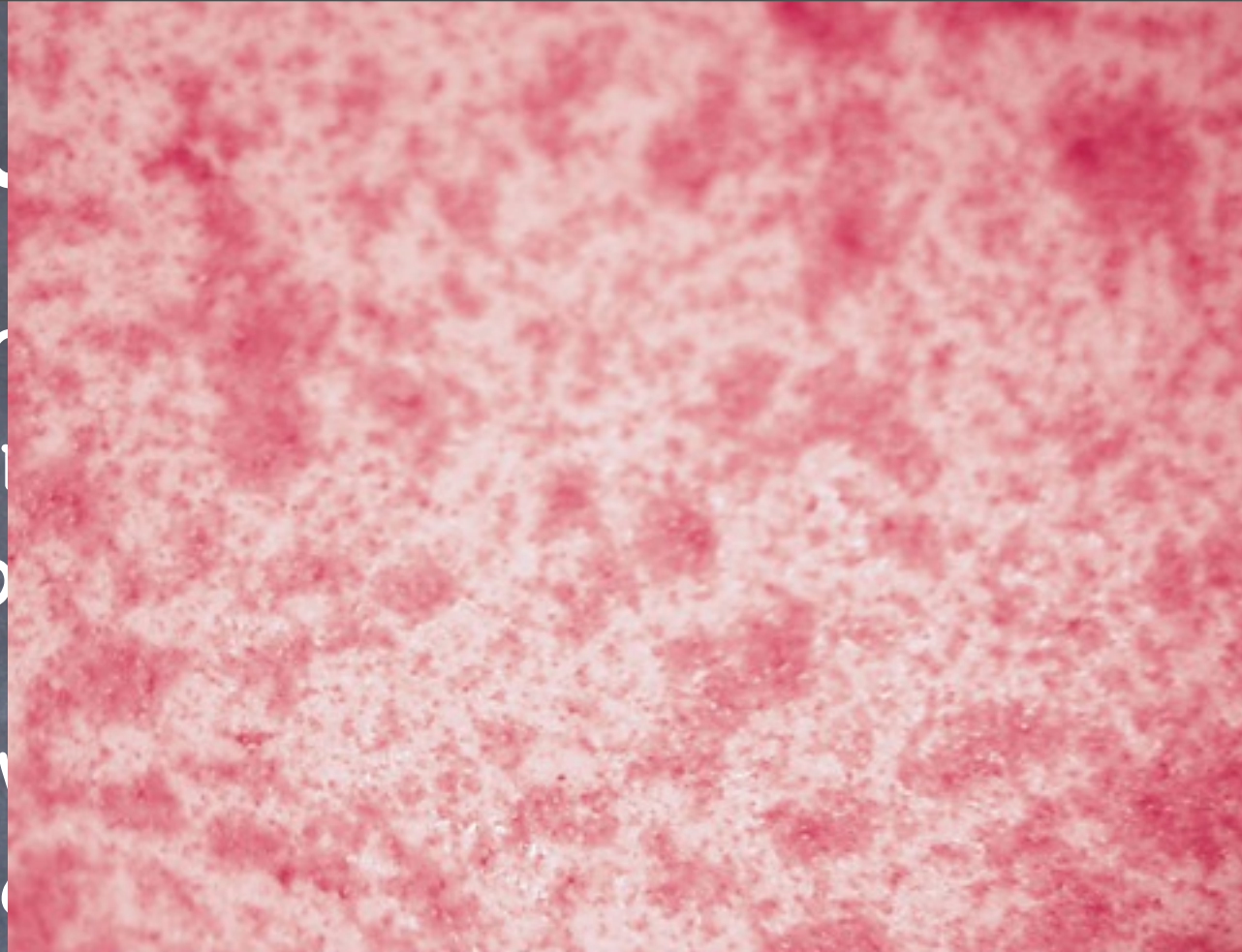
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Natural colours on Earth

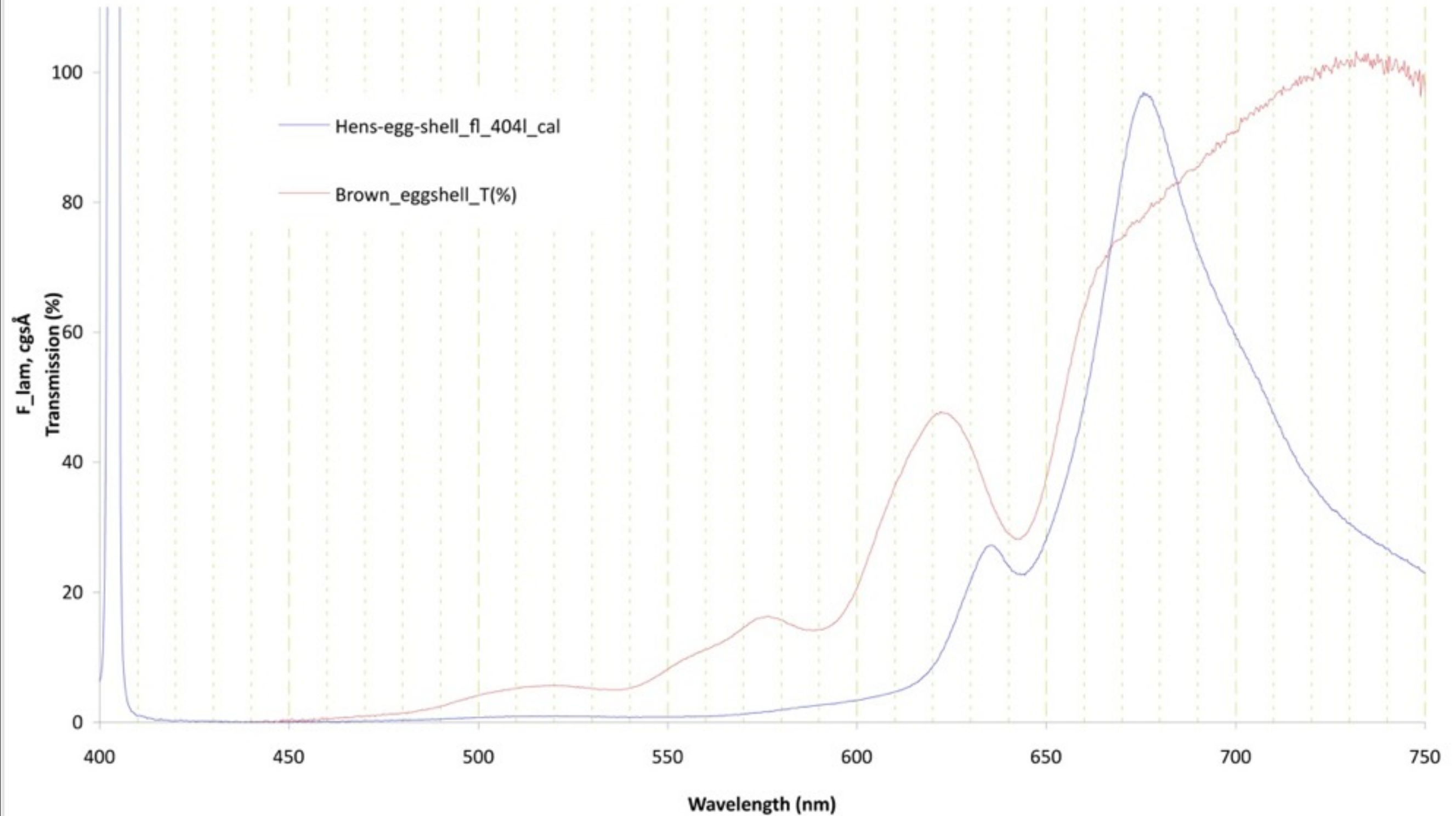
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Natural colours on Earth

Hens egg, shell transmission and fluorescence, 404-laser



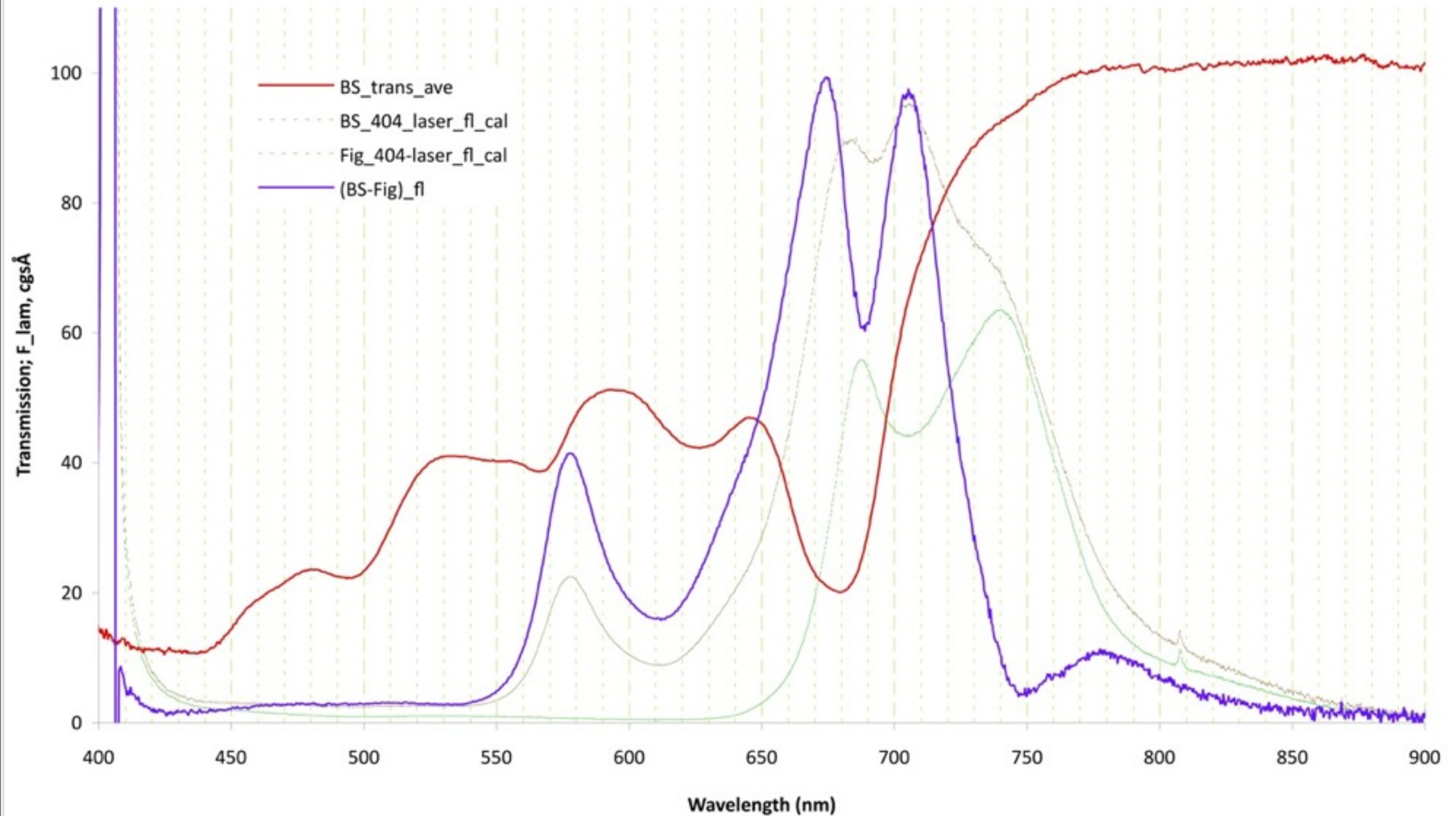
Natural colours on Earth

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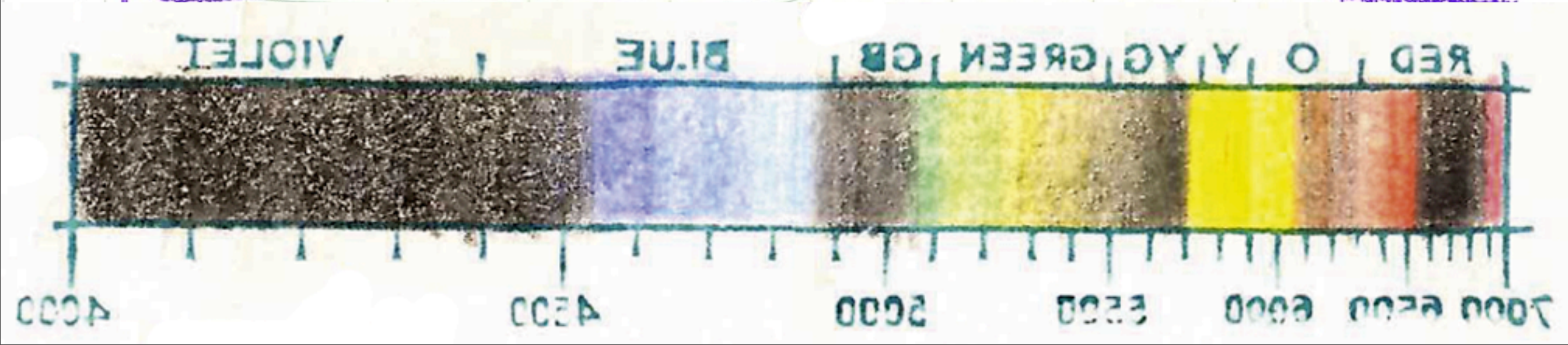
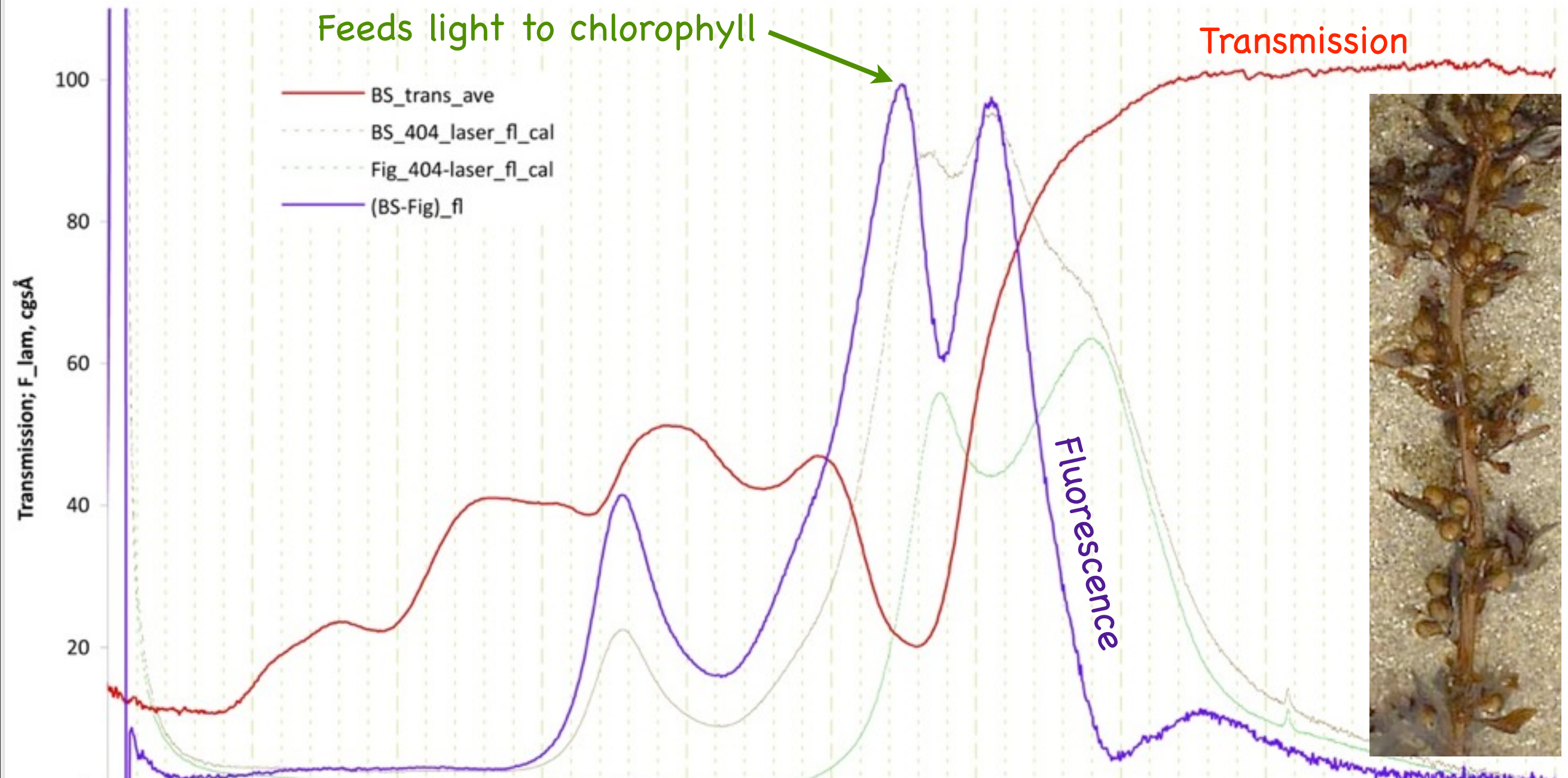
Seaweeds contain auxiliary pigments that transfer energy from underwater (mostly blue!) photons to the photosynthetic mechanisms in chlorophyll

Brown seaweed, Fig, fl and trans; RAEF 11 December 2010



Brown seaweed with phycoerythrin and chlorophyll

Brown seaweed, Fig, fl and trans; RAEF 11 December 2010



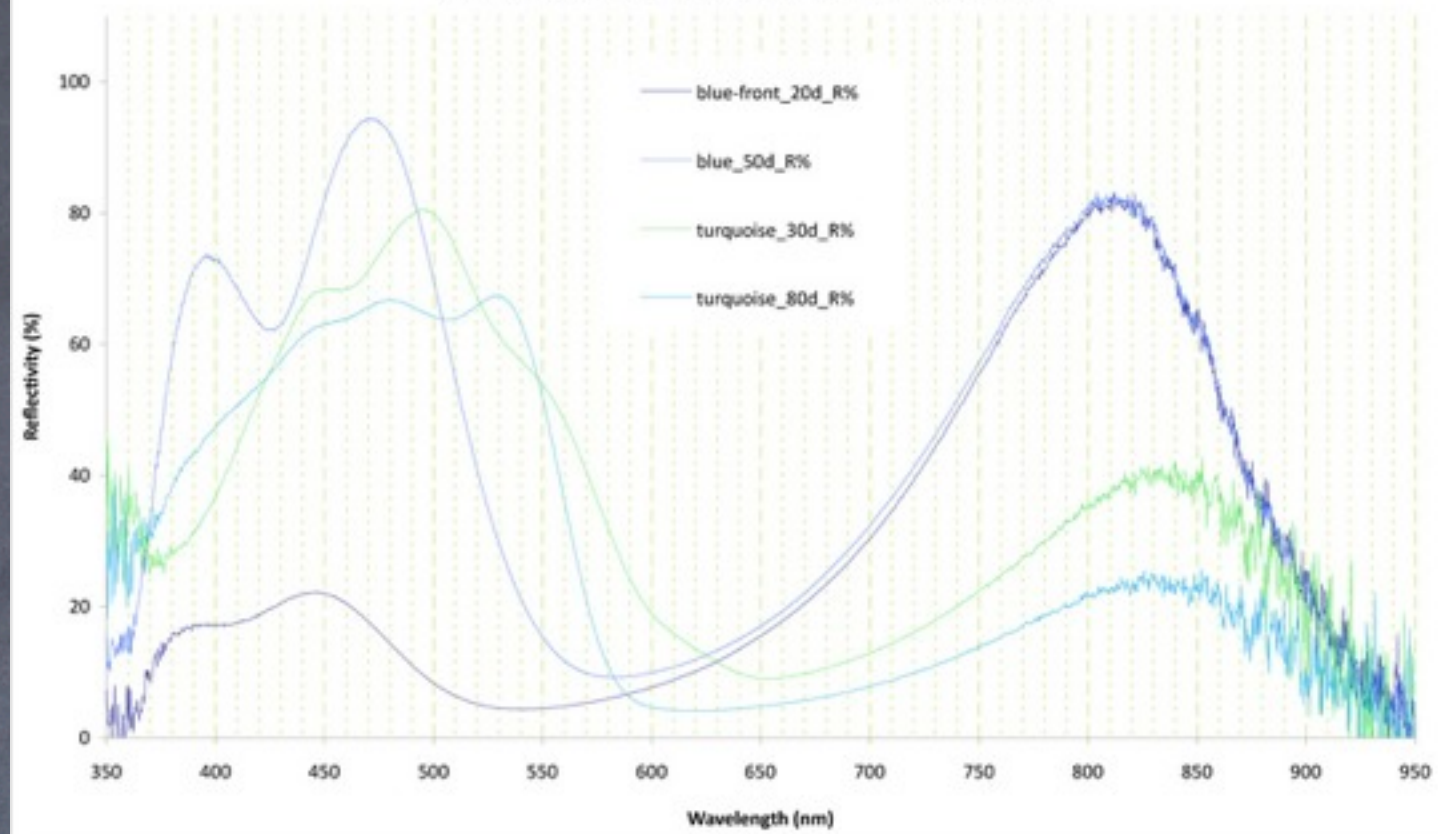
Animal blues

Many animal blues are not caused by pigments but are 'structural' — basically Tyndall (Rayleigh) scattering and/or interference phenomena

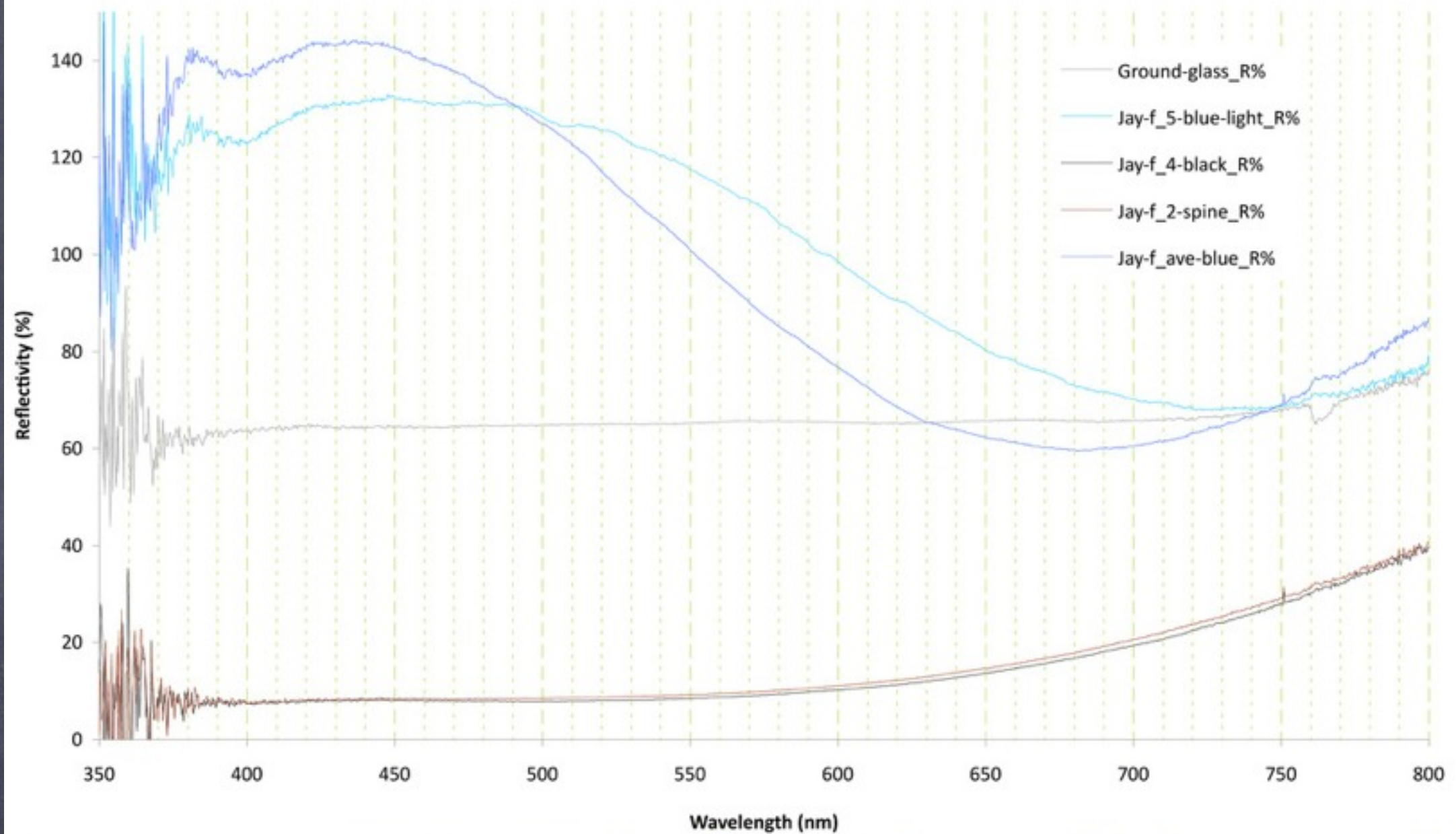
Animal blues



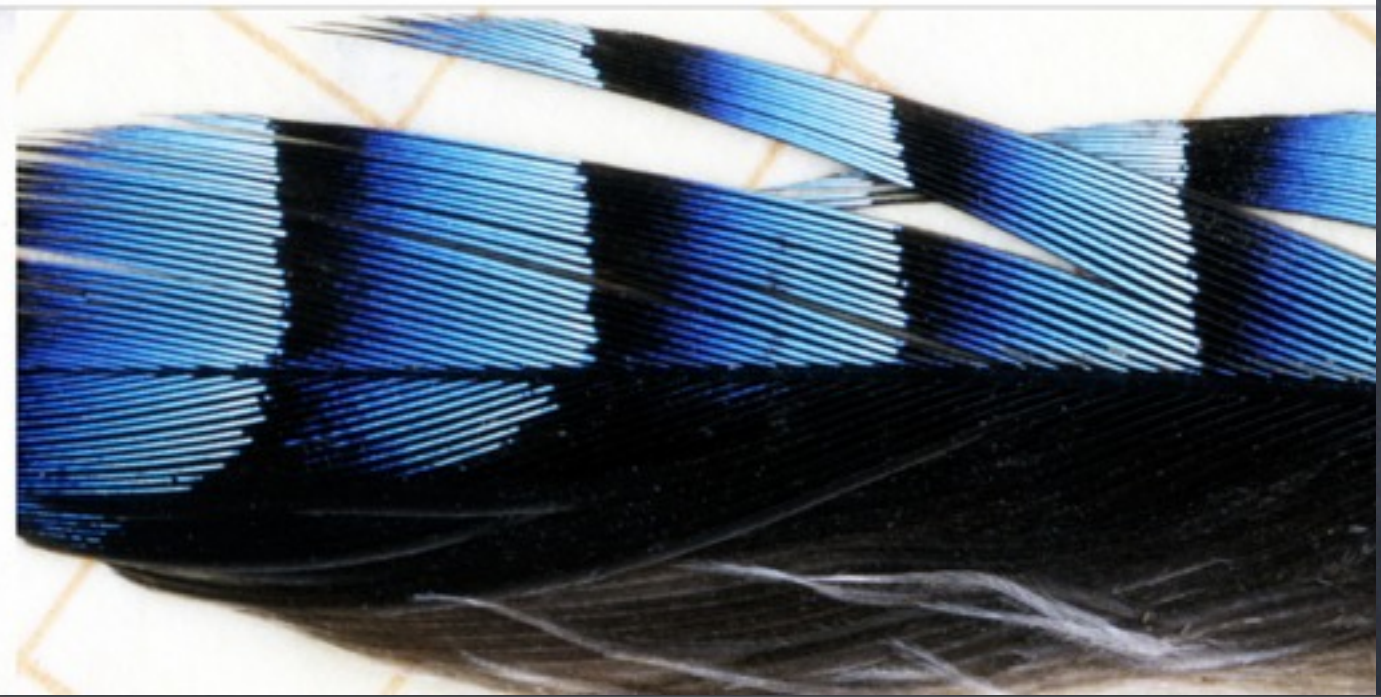
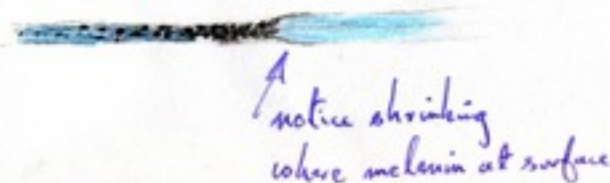
Blue butterfly wing scales, Reflectivity; RAEF 27 March 2011



Jay's feather reflectivity; RAEF 26 March 2011



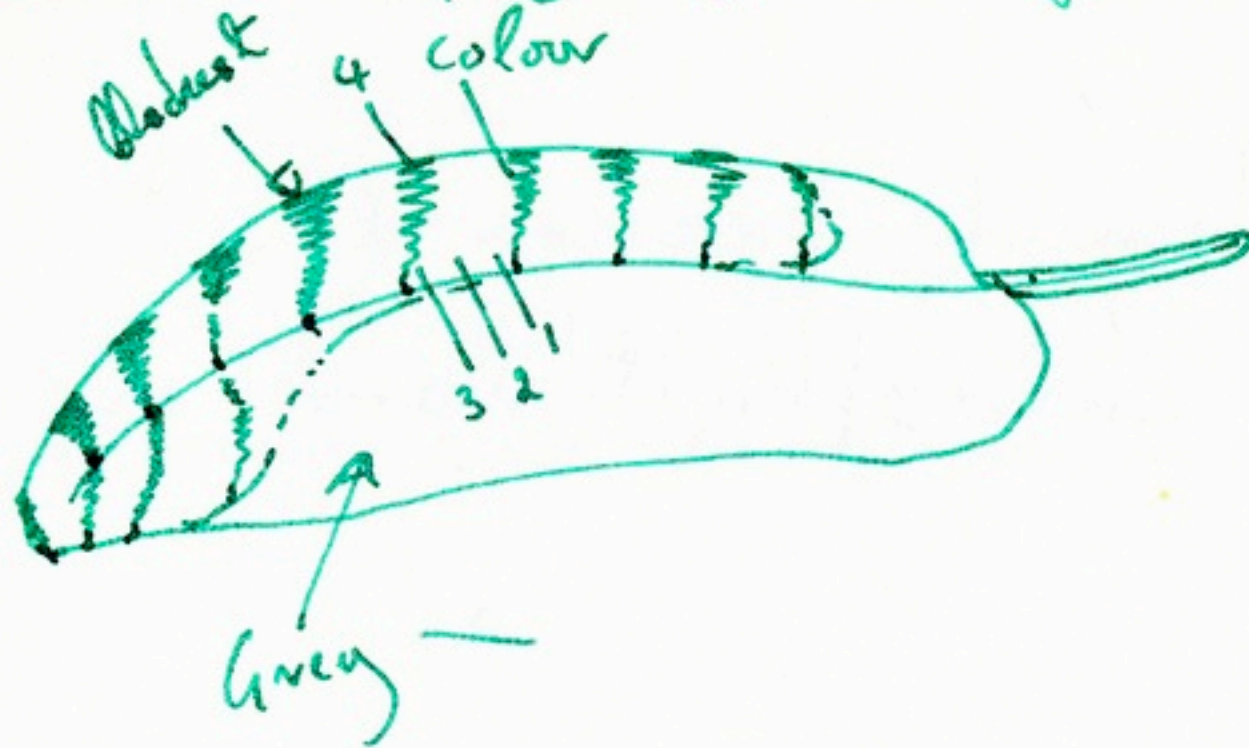
RAEF
29/2/24



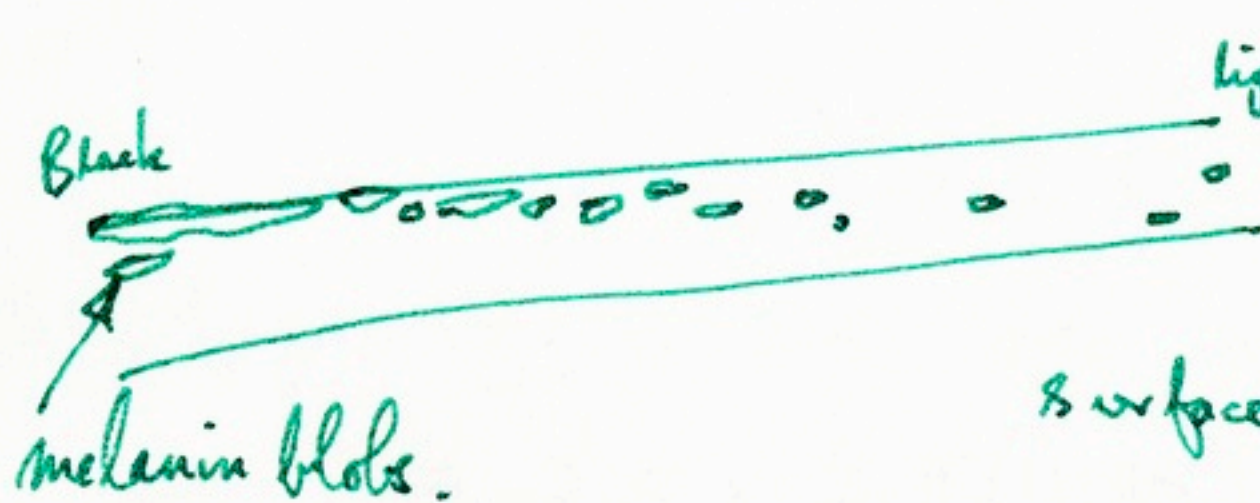
13/9/81

Lay's feather found by Richard Bingham.

99



1. light sky-blue
2. medium blue
3. dark blue
4. tar-black.



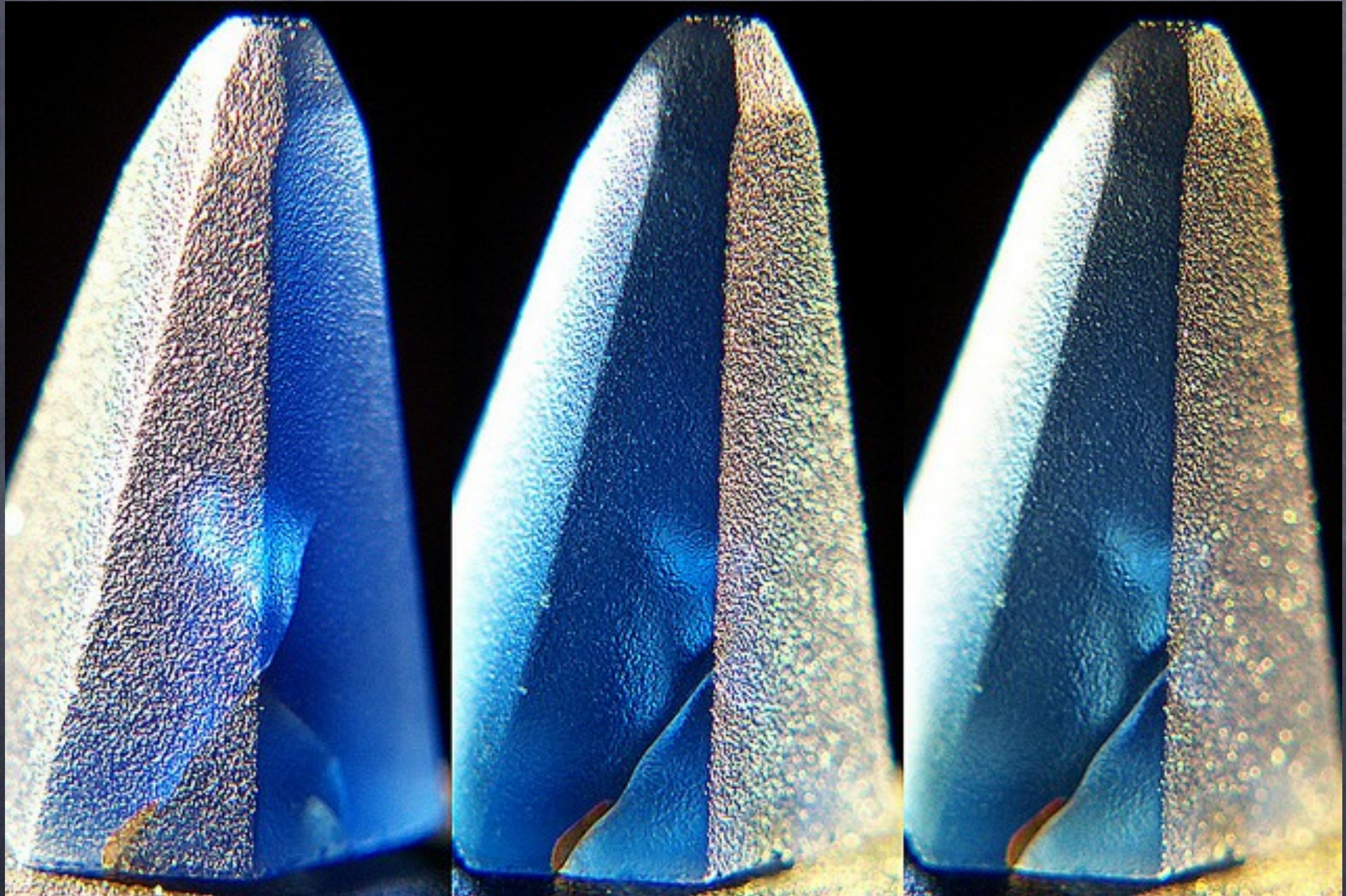
Uniform Tyndall scattering medium. Gradation of colour caused by increase near surface of black melanin blobs.

RACF
29/2/84

A notice shrinking
where melanin at surface



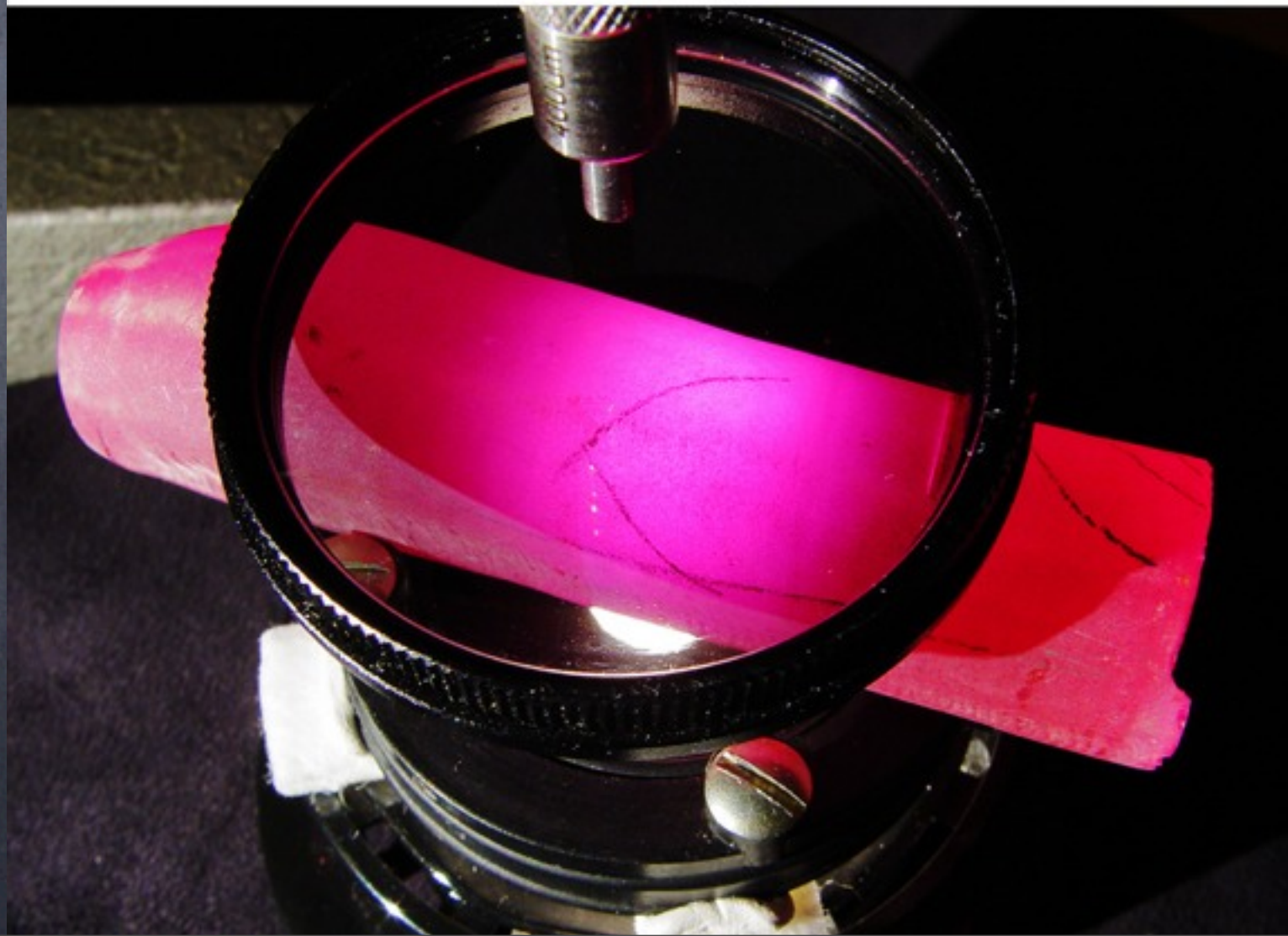
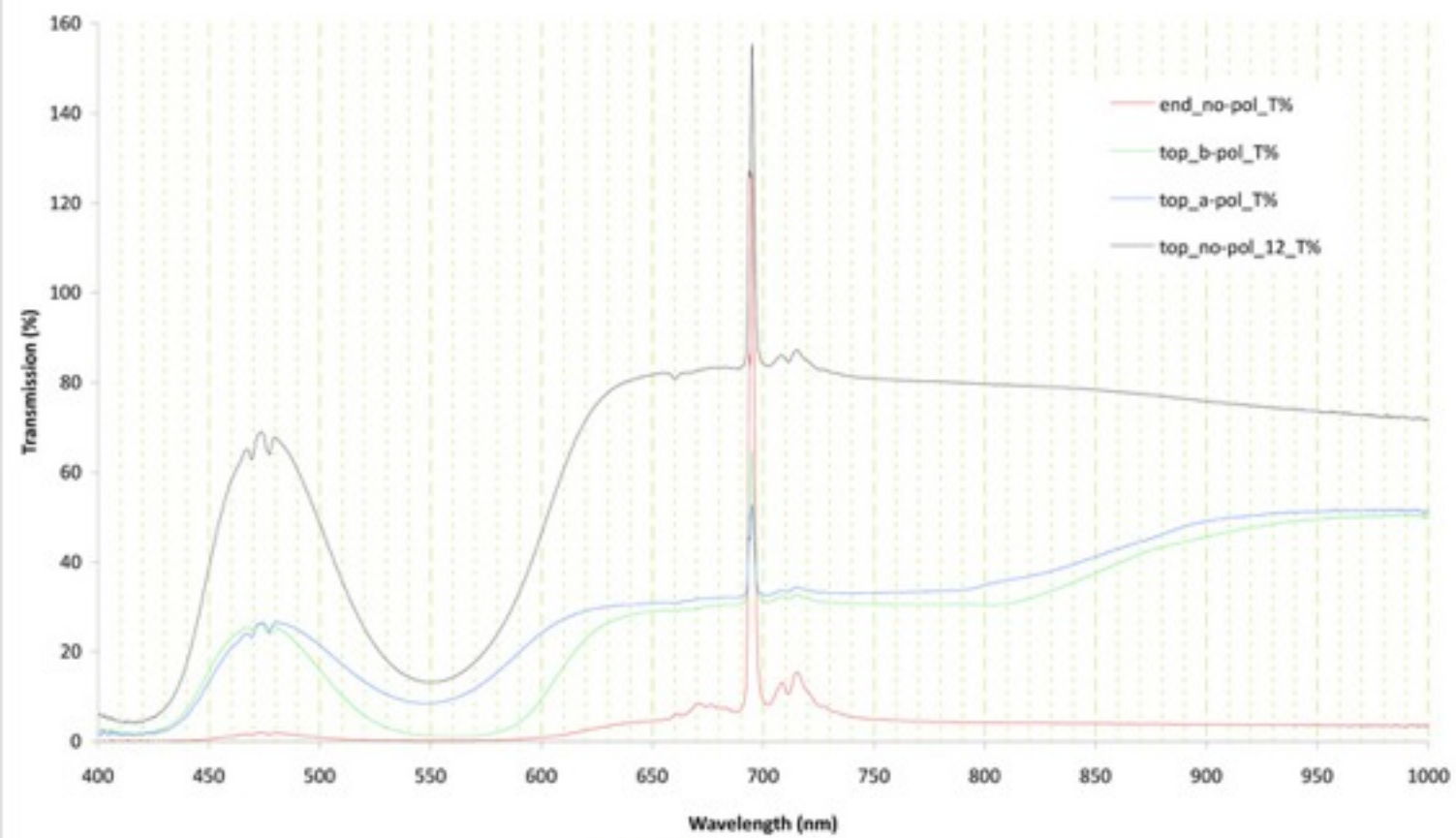
Solids (gemstones)



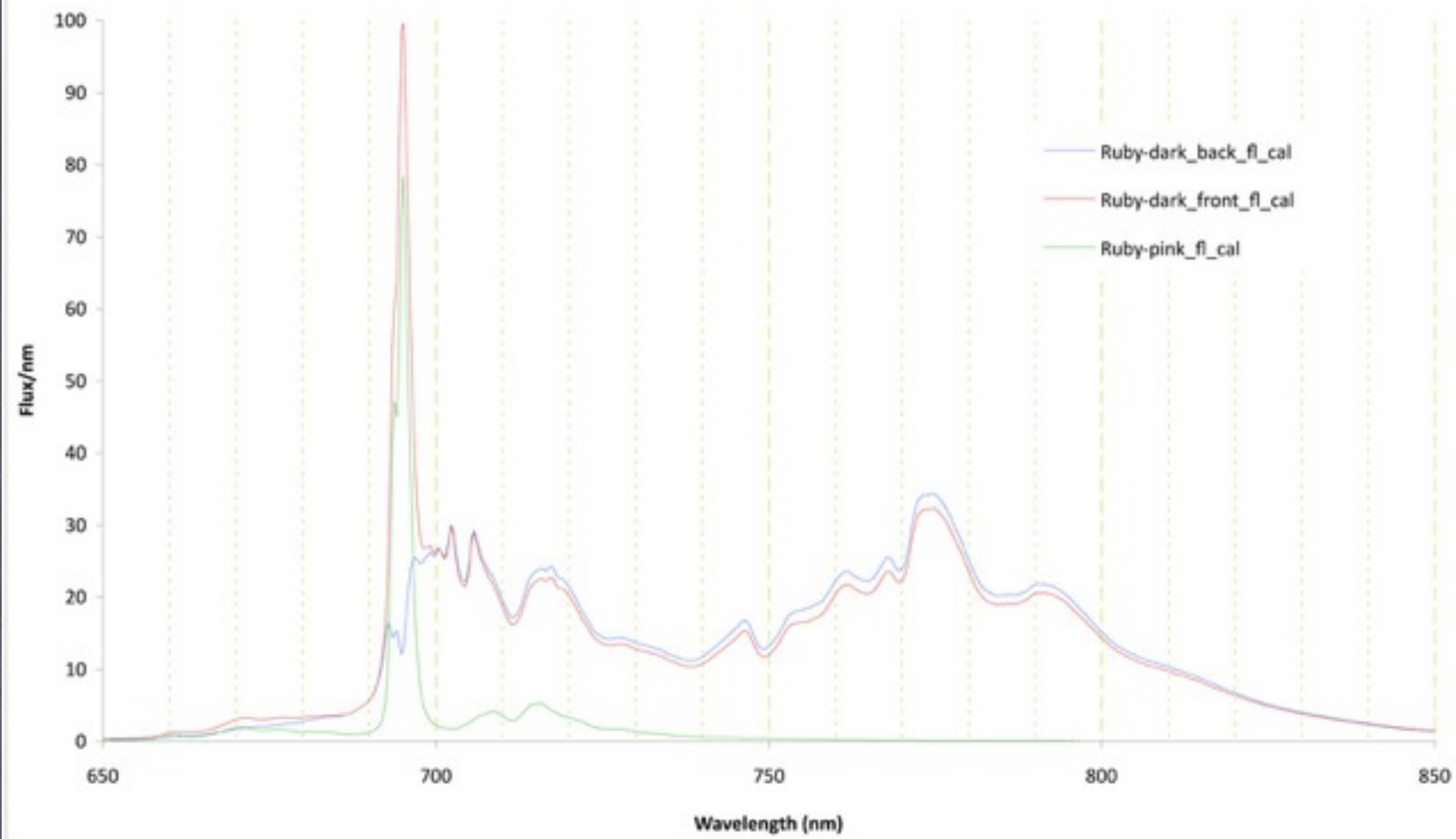
Solids (gemstones)

Different mechanisms produce colour in solids — sometimes this is intrinsic to the basic material but is often due to 'impurities'. The most famous is the role of impurity chromium in ruby, emerald, spinel, kyanite etc. where the surrounding crystal structure generates a ligand field that allows optical transitions. Transitions from metastable levels can lase (the ruby laser)

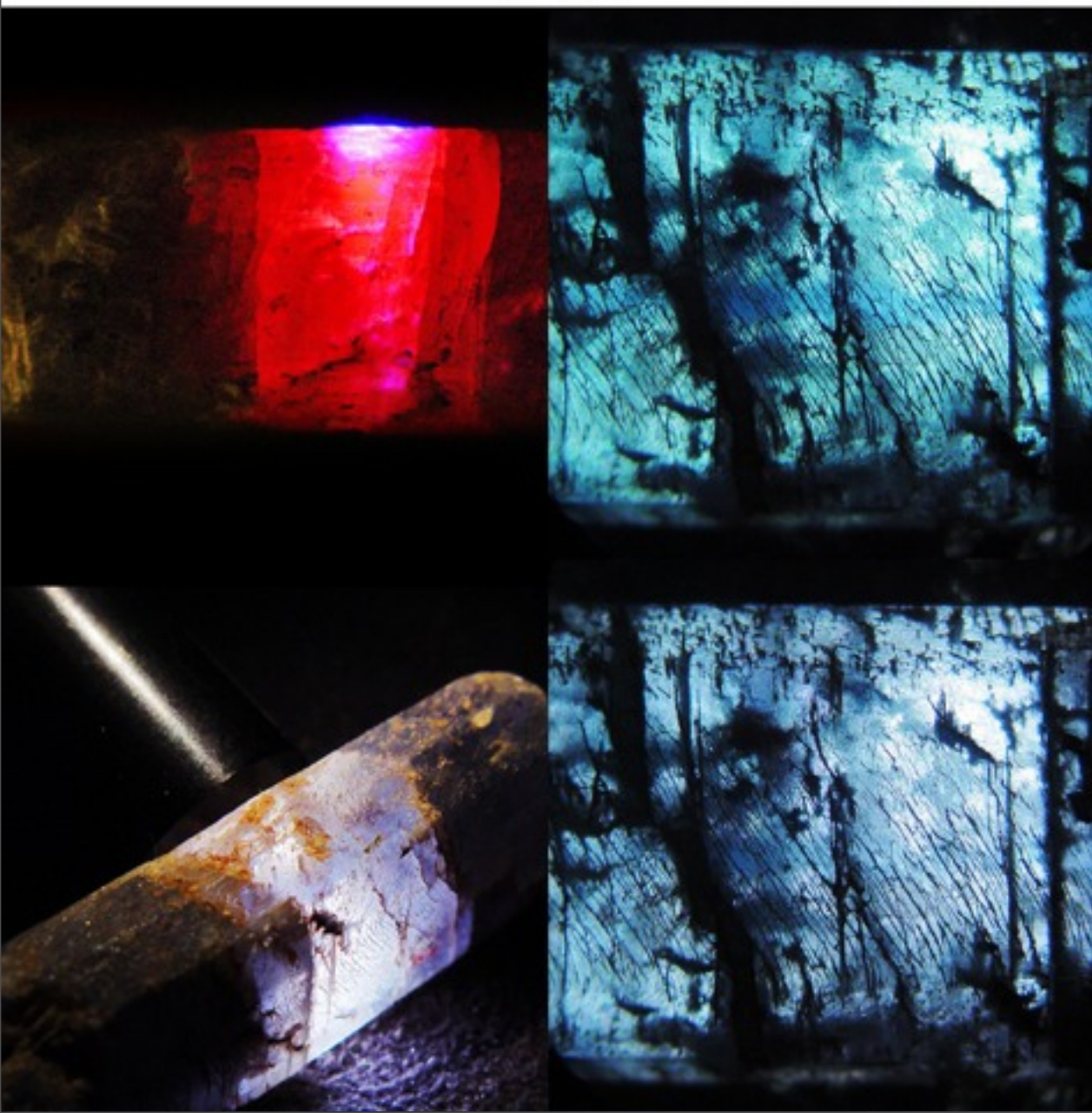
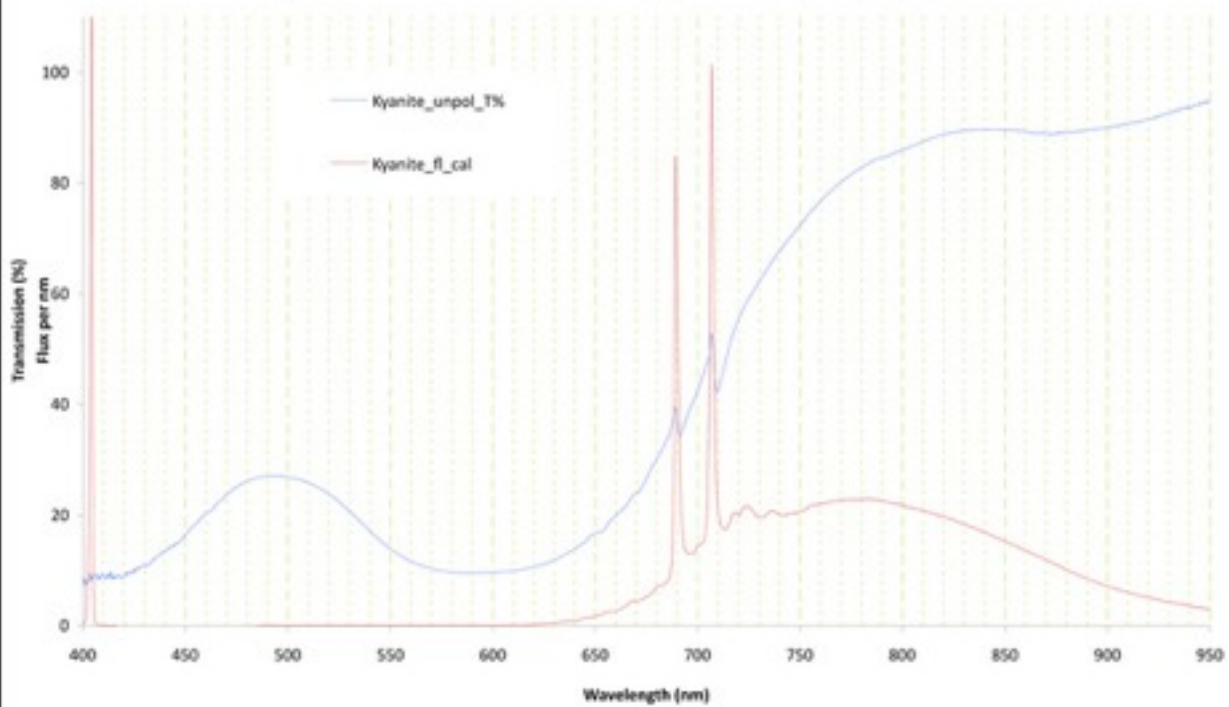
Ruby rod boule; RAEF, 13 April 2011



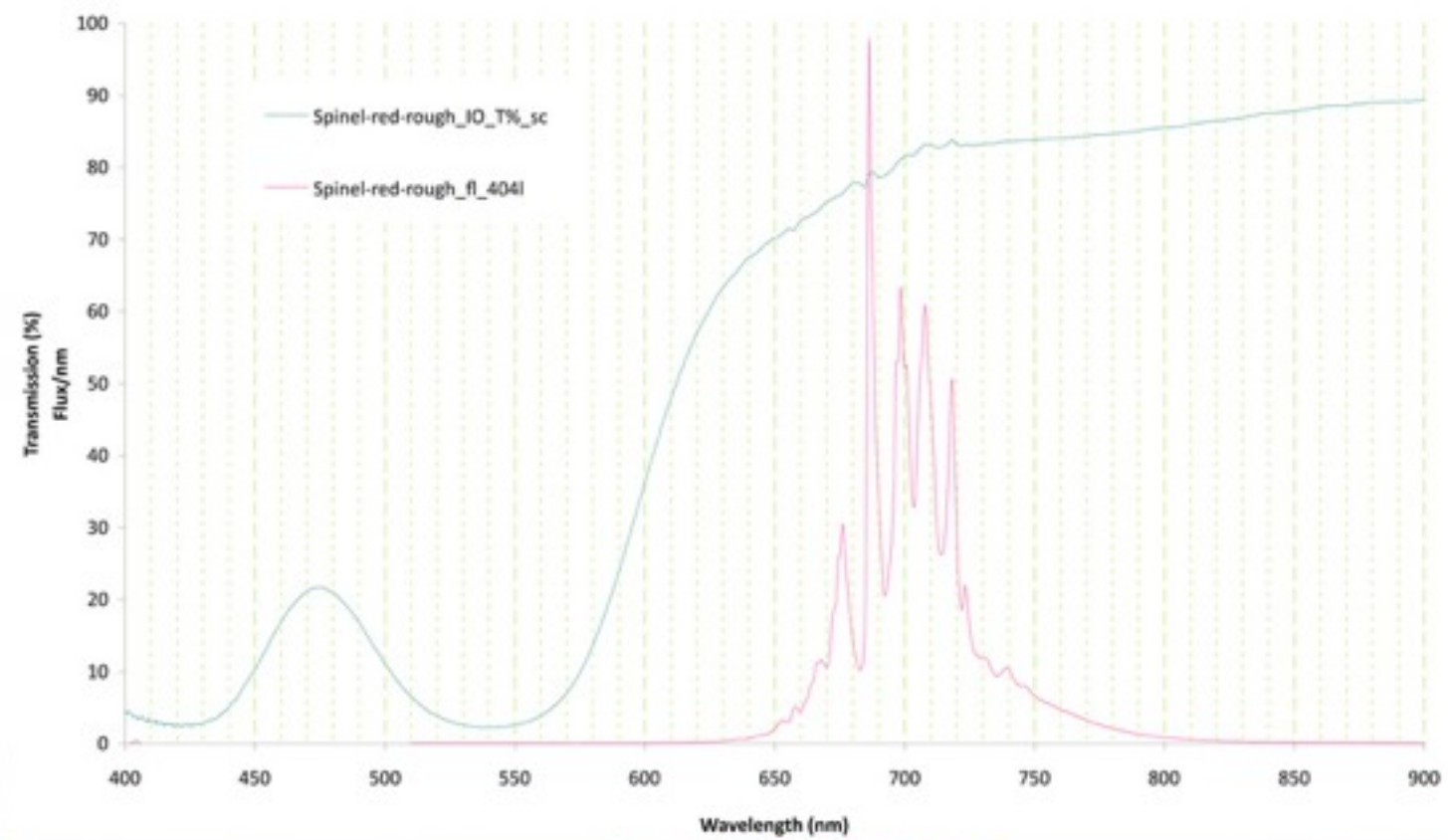
Ruby synthetic cut, pink and deep red; fluorescence, 404nm laser ex; RAEF, 29 April 2011



Kyanite transmission and fluorescence (404nm laser); RAEF, 16 April 2011



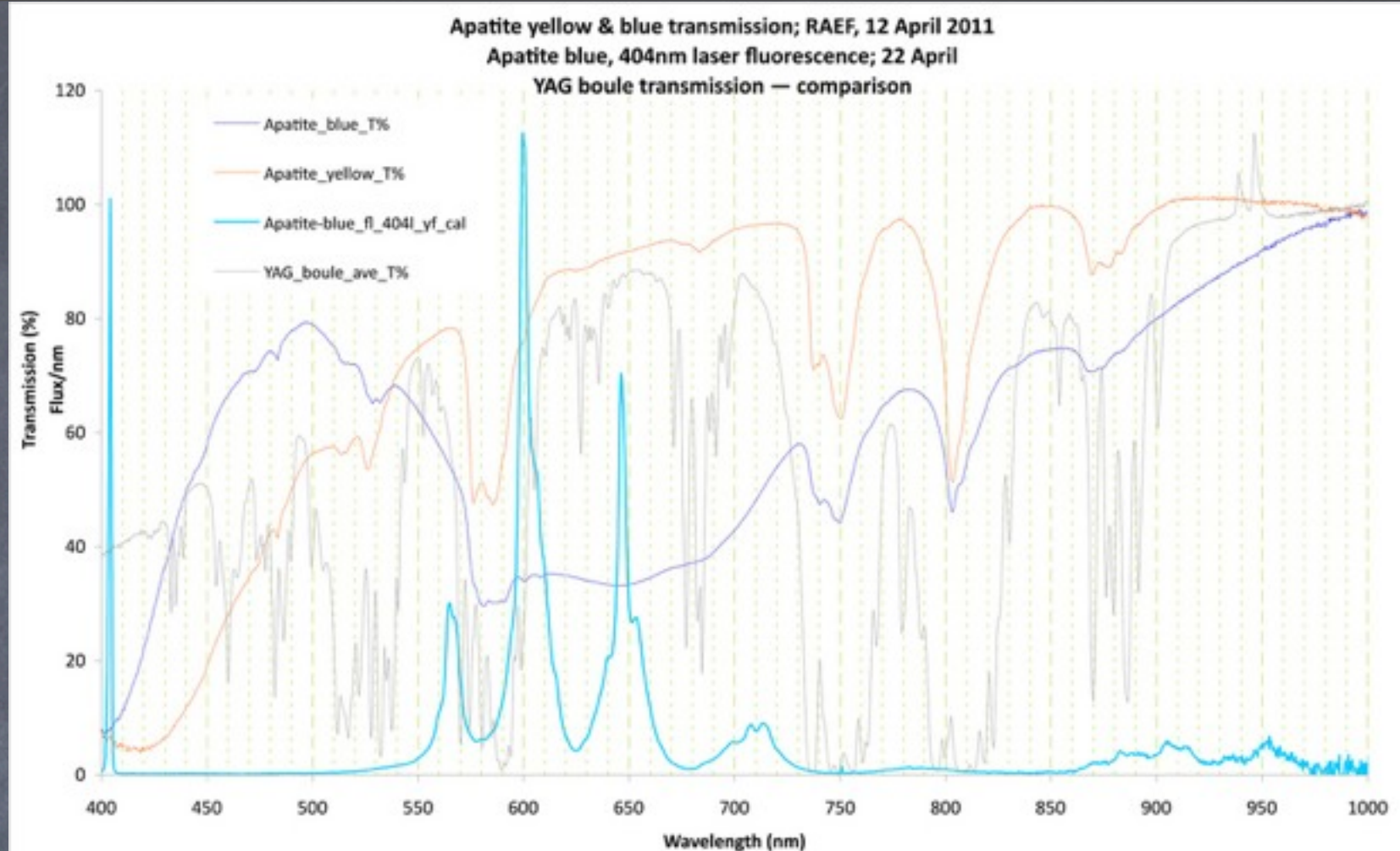
Spinel red rough, transmission and 404nm laser fluorescence; RAEF, 27 April 2011



Apatite (calcium phosphate – Chloride Fluoride Hydroxide) contains rare earths (eg. neodymium) that make inner-shell transitions and produce rich absorption spectra. The Nd^{3+} emits the strong NIR fluorescence lines that are exploited in Nd YAG lasers

The transmission spectrum of YAG is shown as the faint grey line

There is lots of apatite in your bones and teeth



The effect of decades of desert sunlight on glass containing manganese and iron

This old glass telephone insulator was found on a fallen (unused!) telegraph pole by Trish on a trip from Parkes to Sydney

The colour (purple with high transparency in the blue) is from UV-induced colour-centres similar to those in adaptive spectacles



Ravenala Madagascariensis
A 'new' blue



Ravenala Madagascariensis
A 'new' blue



Ravenala Madagascariensis

A 'new' blue

extended blue sensitivity



Pl. 4.



Half ant.

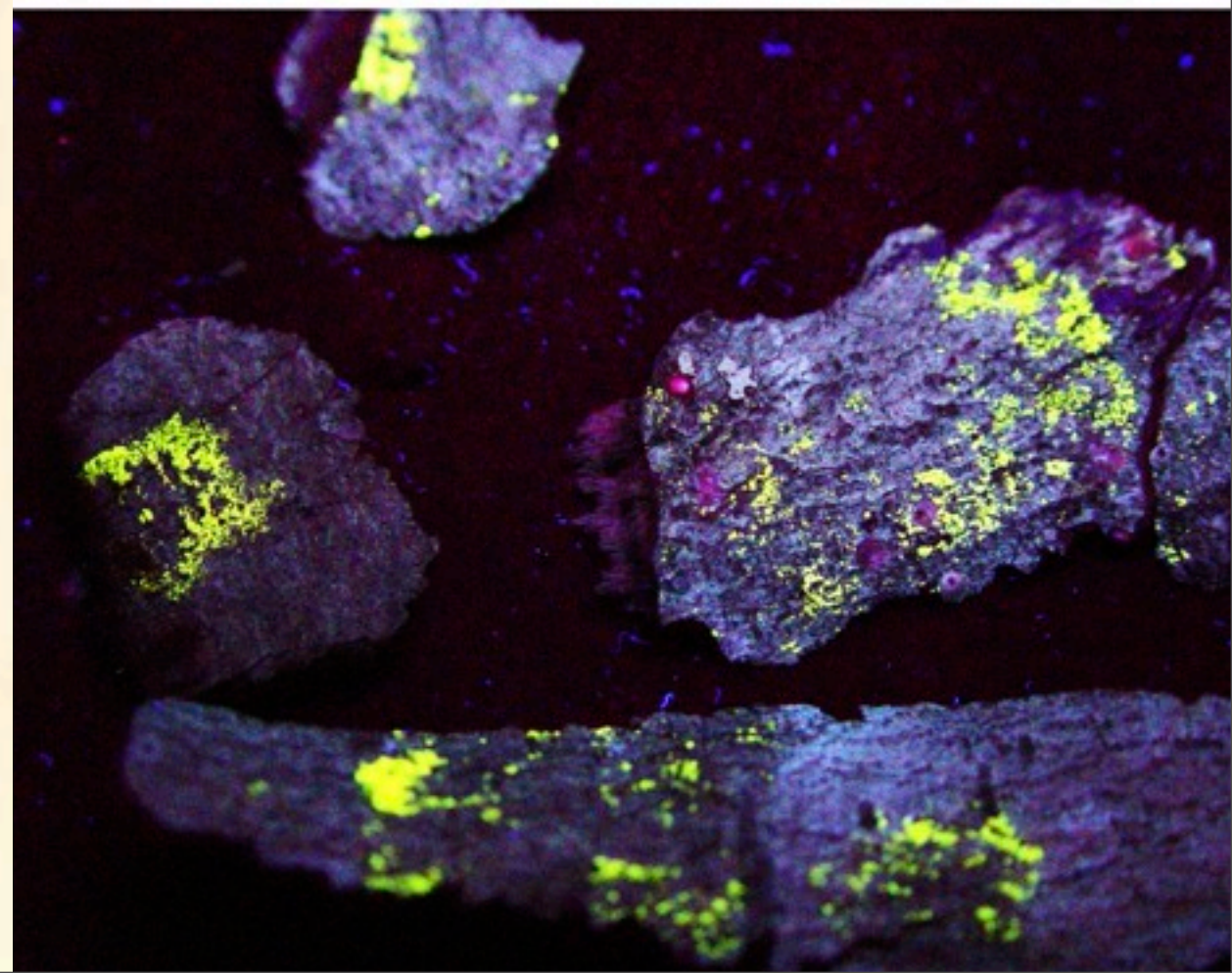
Thomson, Phil. Soc. T. 1865. Pl. 70115.



Pl. 4.



Thomson, Phil. Soc. T. 1865. Pl. 7000.



The colours discussed in the first part of the talk are (and will be) important for characterising exo-Earths. It may be a while, however, before the later examples become of interest to astrophysicists (except to me of course!)

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